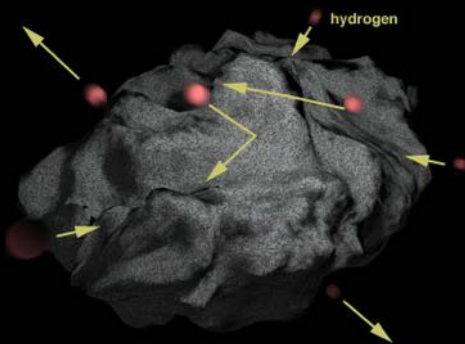
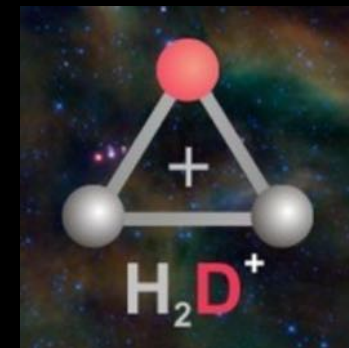
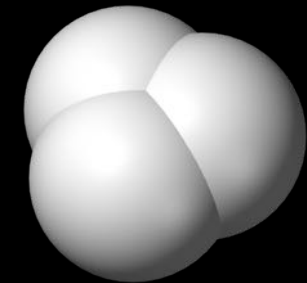


1

Astrochemistry with SOFIA



Paola Caselli
Center for Astrochemical Studies
Max-Planck-Institute for Extraterrestrial Physics



FISO Tele-Talk 1 June 2016

2

Molecular clouds in the Milky Way

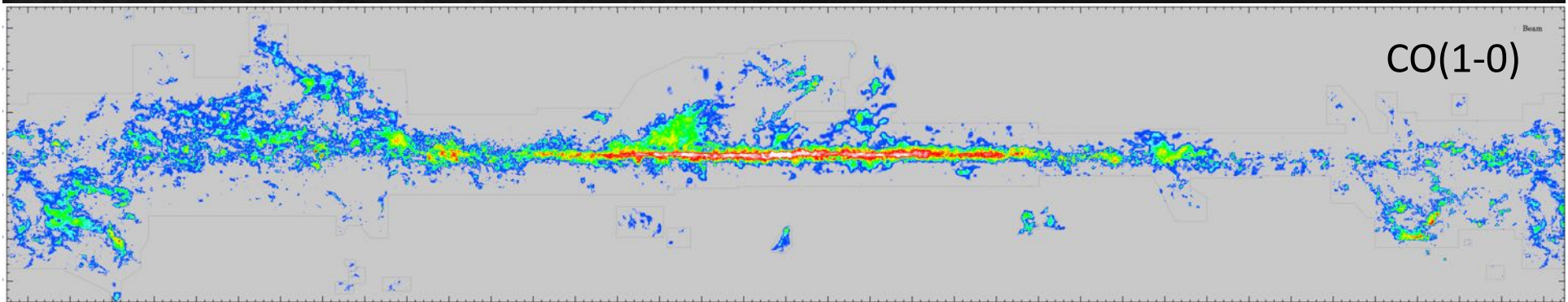
~ 100,000 light years



3

Molecular clouds in the Milky Way

~ 100,000 light years



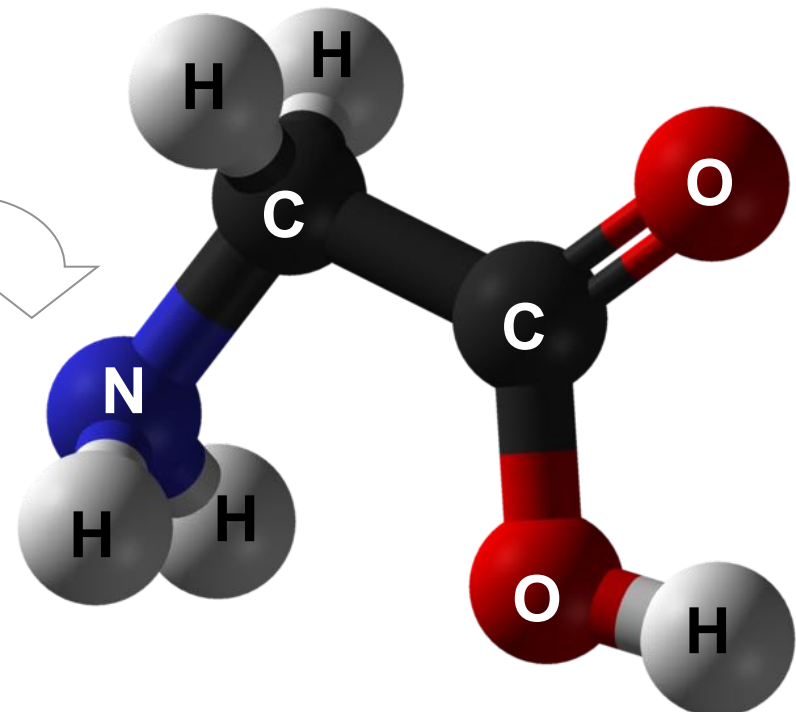
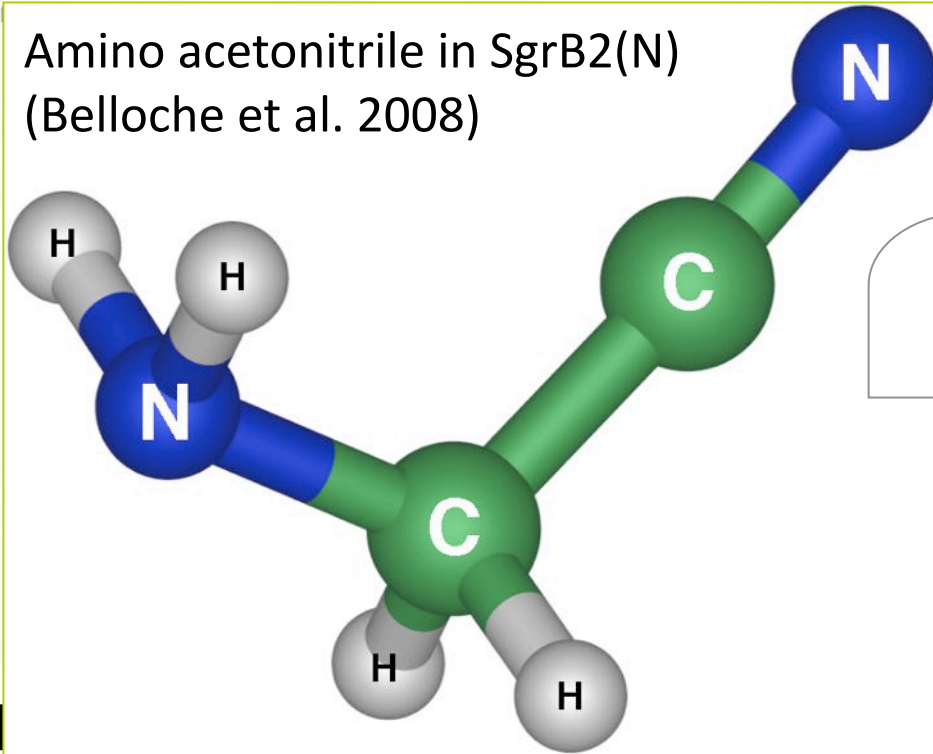
Dame, Hartmann & Thaddeus 2001

5 Interstellar molecules and the seeds of life

Known Interstellar Molecules (Total ~200)

Number of Atoms									
2	3	4	5	6	7	8	9		
H ₂	H ₂ O	NH ₃	SiH ₄	CH ₃ OH	CH ₃ COH	CH ₃ CO ₂ H	CH ₃ CH ₂ OH		
OH	H ₂ S	H ₃ O ⁺	CH ₄	NH ₂ CHO	CH ₃ NH ₂	HCO ₂ CH ₃	(CH ₃) ₂ O		
SO	SO ₂	H ₂ CO	CHOOH	CH ₃ CN	CH ₃ CCH	CH ₃ C ₂ CN	CH ₃ CH ₂ CN		
SO ⁺	HN ₂ ⁺	H ₂ CS	HCCCN	CH ₃ NC	CH ₂ CHCN	C ₇ H	H(CC) ₃ CN		
SiO	HNO	HNCO	CH ₂ NH	CH ₃ SH	HC ₄ CN	H ₂ C ₆	H(CC) ₂ CH ₃		
SiS	SiH ₂ ?	HNCS	NH ₂ CN	C ₅ H	C ₆ H	C ₇	C ₈ H		

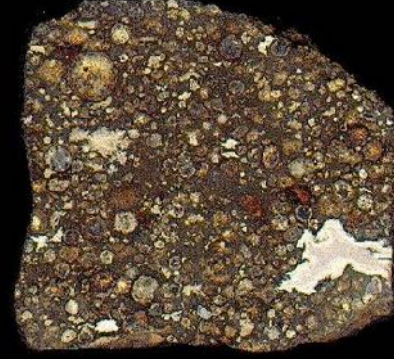
Amino acetonitrile in SgrB2(N)
(Belloche et al. 2008)



Glycine - the simplest amino acid

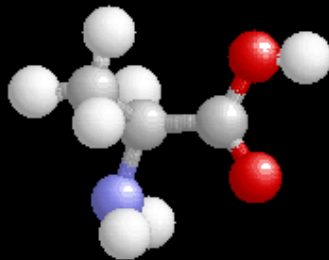
6

Pre-biotic molecules in meteorites

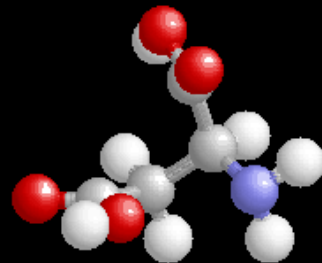


ALLENDE, CV3, MEXICO
Photo & Collection
Harald Seehlik

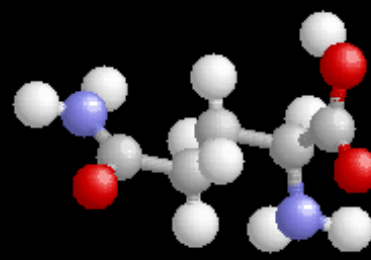
**~70 amino acids have been identified in meteorites;
21 of these are used in life.**



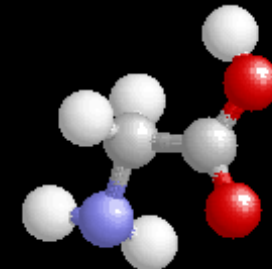
L-Alanine



L-Aspartic Acid



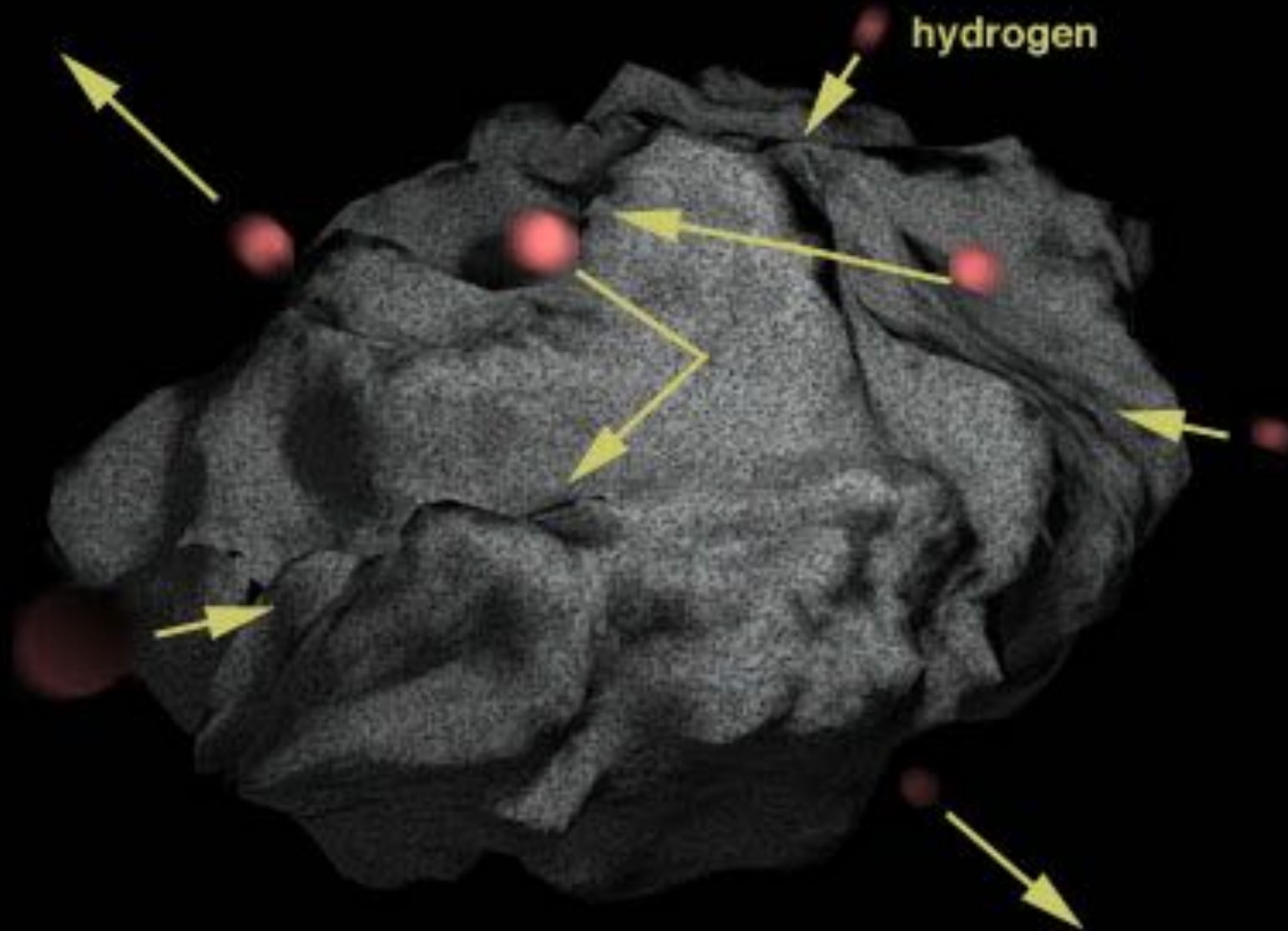
L-Glutamine



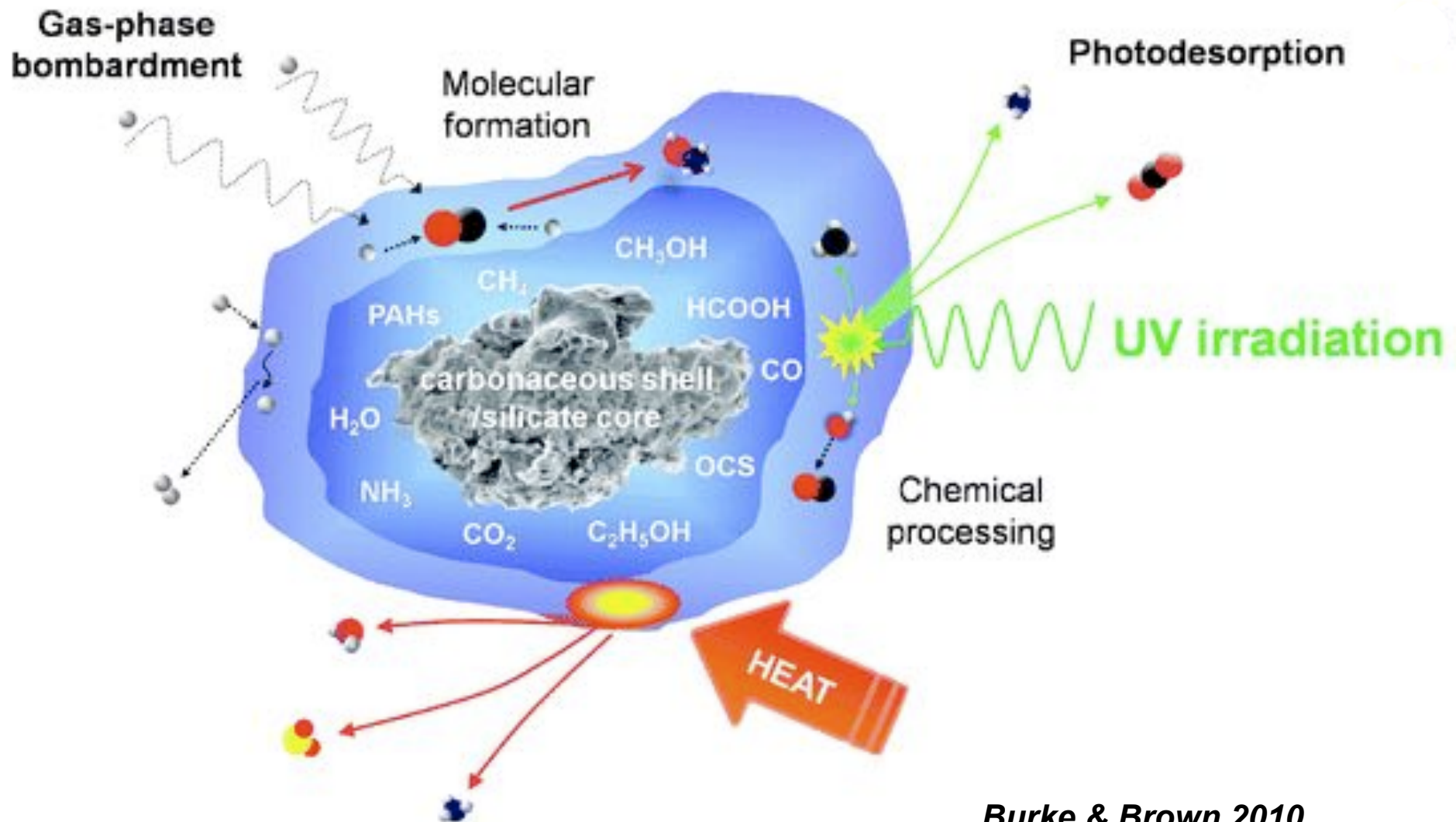
Glycine

7

Interstellar molecules and the seeds of life



8 Interstellar molecules and the seeds of life

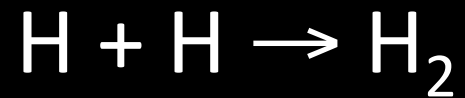


Burke & Brown 2010

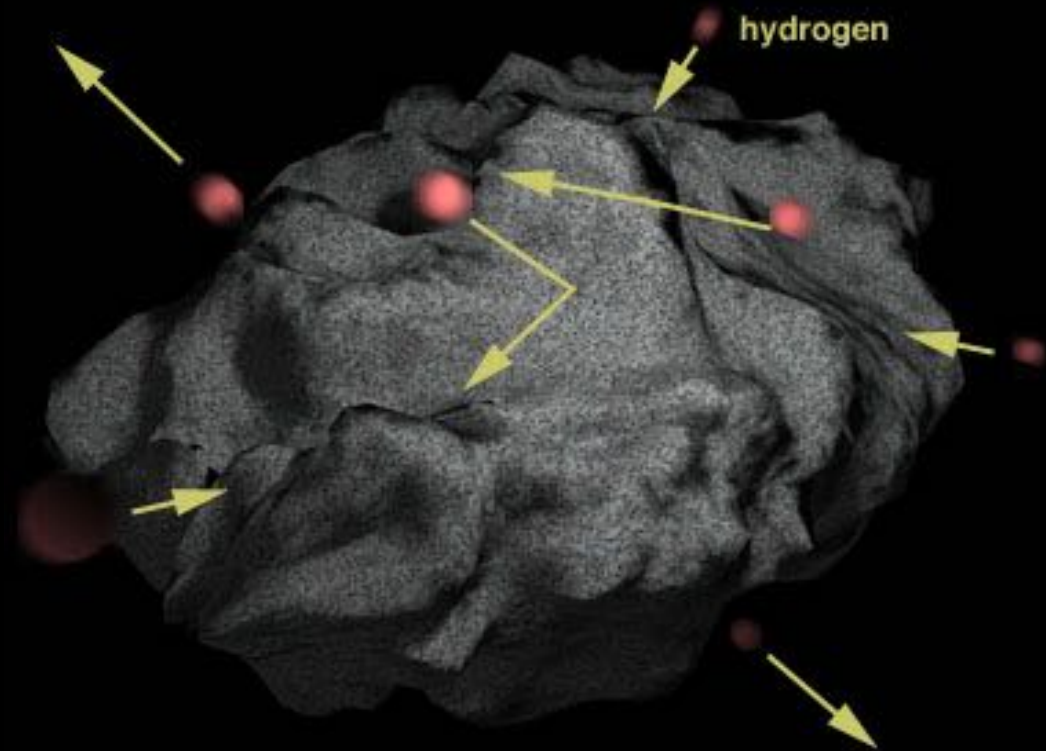
- Formation of H_2
- Formation of H_3^+
- Deuterium fractionation
- The ortho-to-para H_2 ratio
- SOFIA discovery of para- H_2D^+
- SOFIA and oxygen chemistry
- SOFIA and sulfur chemistry

The formation of H₂

The reaction that starts the chemistry in the interstellar medium is the one between two hydrogen atoms to form molecular hydrogen:



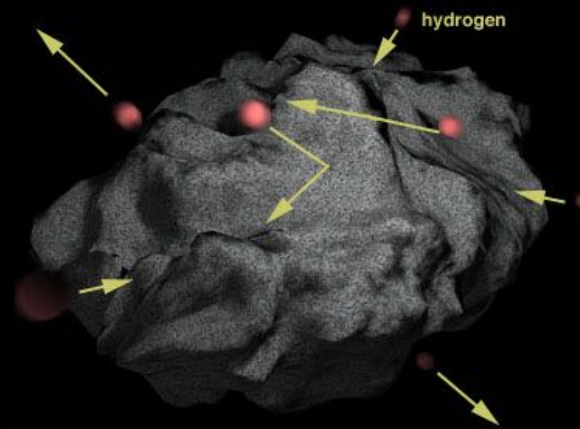
This reaction happens on the surface of dust grains.



11

The H_2 formation rate ($\text{cm}^{-3} \text{s}^{-1}$) is given by (e.g. Gould & Salpeter 1963; Hollenbach & Salpeter 1970; Jura 1974; Pirronello et al. 1999; Cazaux & Tielens 2002; Bergin et al. 2004; Cuppen & Herbst 2005; Cazaux et al. 2008):

$$R_{H_2} = \frac{1}{2} n_H v_H A n_g S_H \gamma$$
$$\cong 10^{-17} - 10^{-16} \text{ cm}^{-3} \text{ s}^{-1}$$



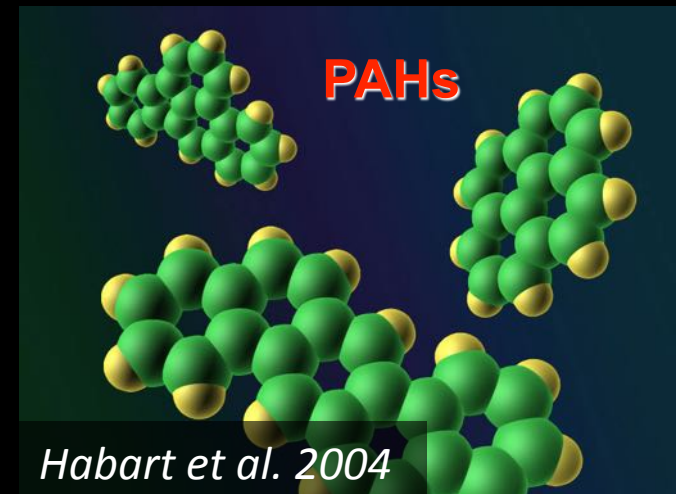
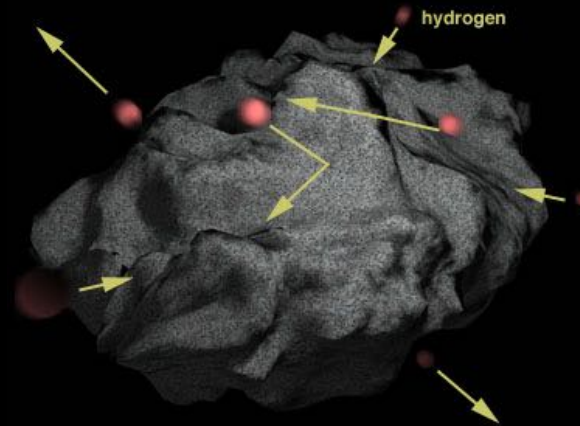
- n_H \equiv gas number density
- v_H \equiv H atoms speed in gas-phase
- A \equiv grain cross sectional area
- n_g \equiv dust grain number density
- S_H \equiv sticking probability
- γ \equiv surface reaction probability

12

The H_2 formation rate ($\text{cm}^{-3} \text{s}^{-1}$) is given by (e.g. Gould & Salpeter 1963; Hollenbach & Salpeter 1970; Jura 1974; Pirronello et al. 1999; Cazaux & Tielens 2002; Bergin et al. 2004; Cuppen & Herbst 2005; Cazaux et al. 2008):

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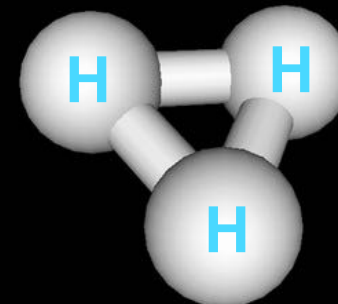
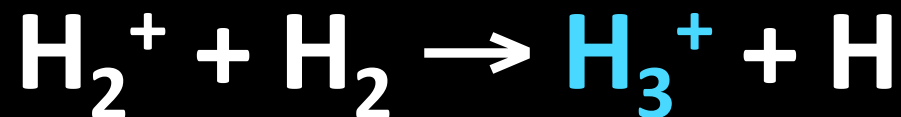


The formation of H_3^+

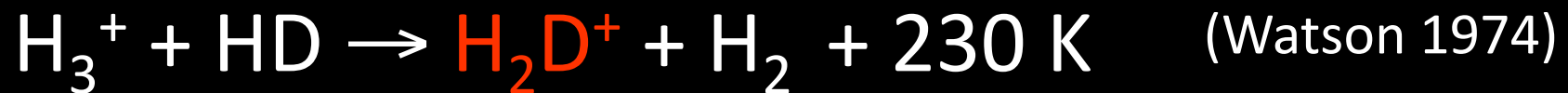
After the formation of molecular hydrogen, **cosmic rays** ionize H_2 initiating fast routes towards the formation of complex molecules in dark clouds:



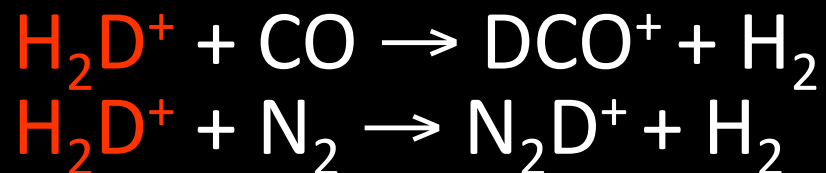
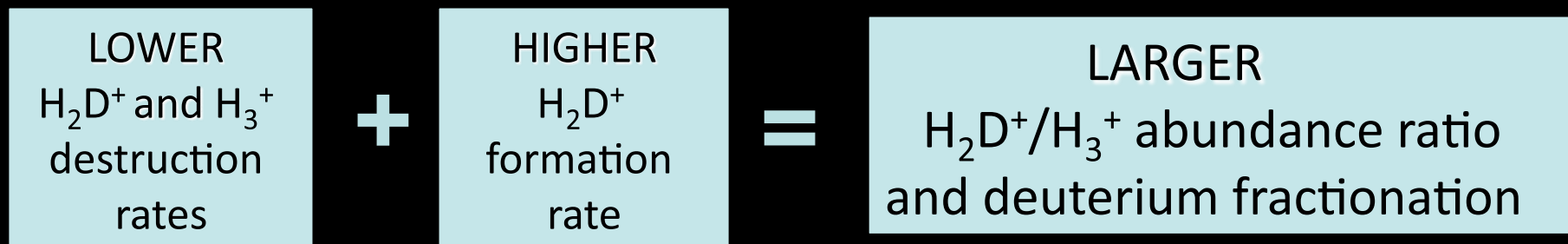
Once H_2^+ is formed (in small percentages), it very quickly reacts with the abundant H_2 molecules to form H_3^+ , the most important molecular ion in interstellar chemistry:



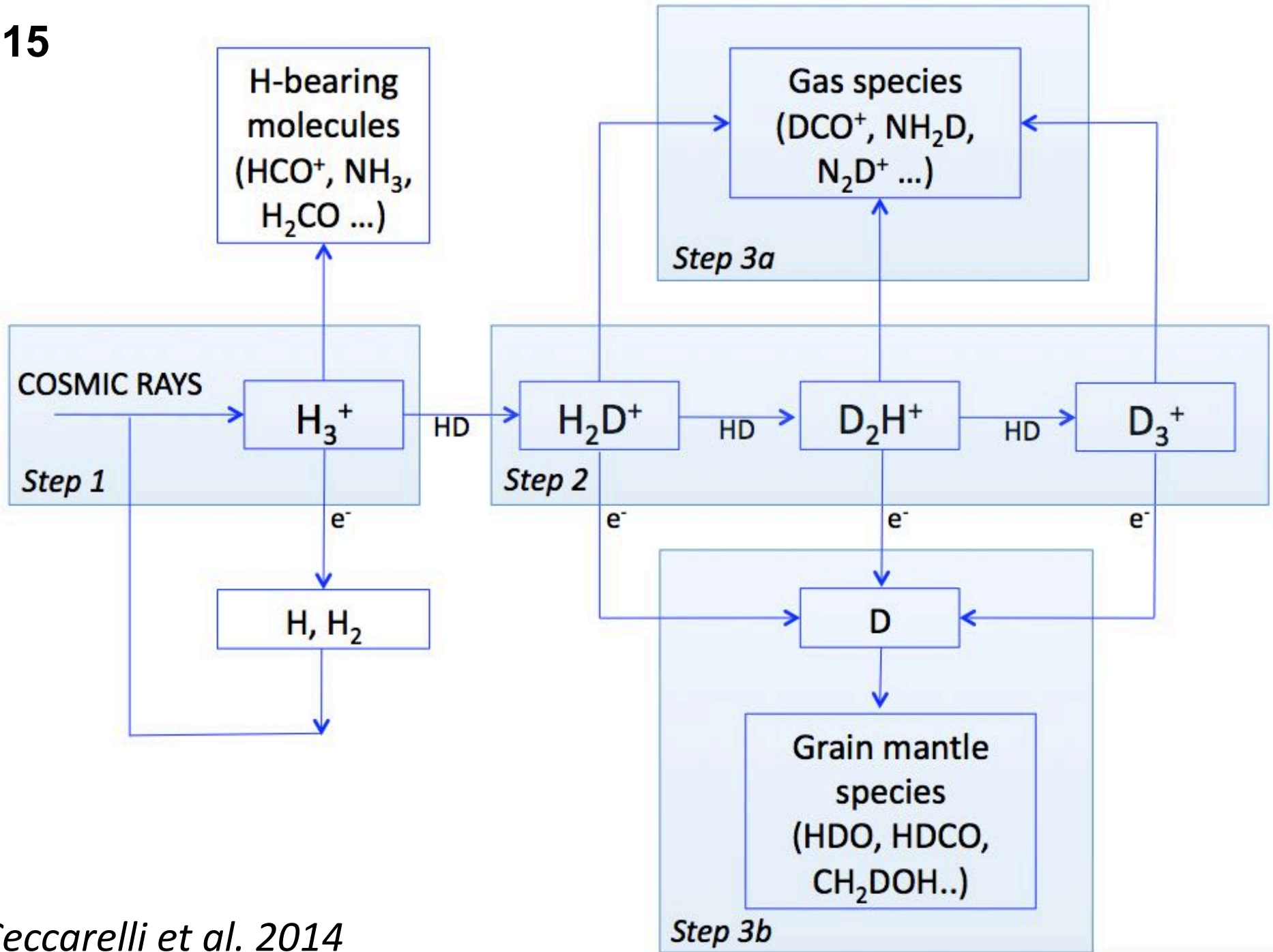
14 Deuterium Fractionation at $T < 20$ K



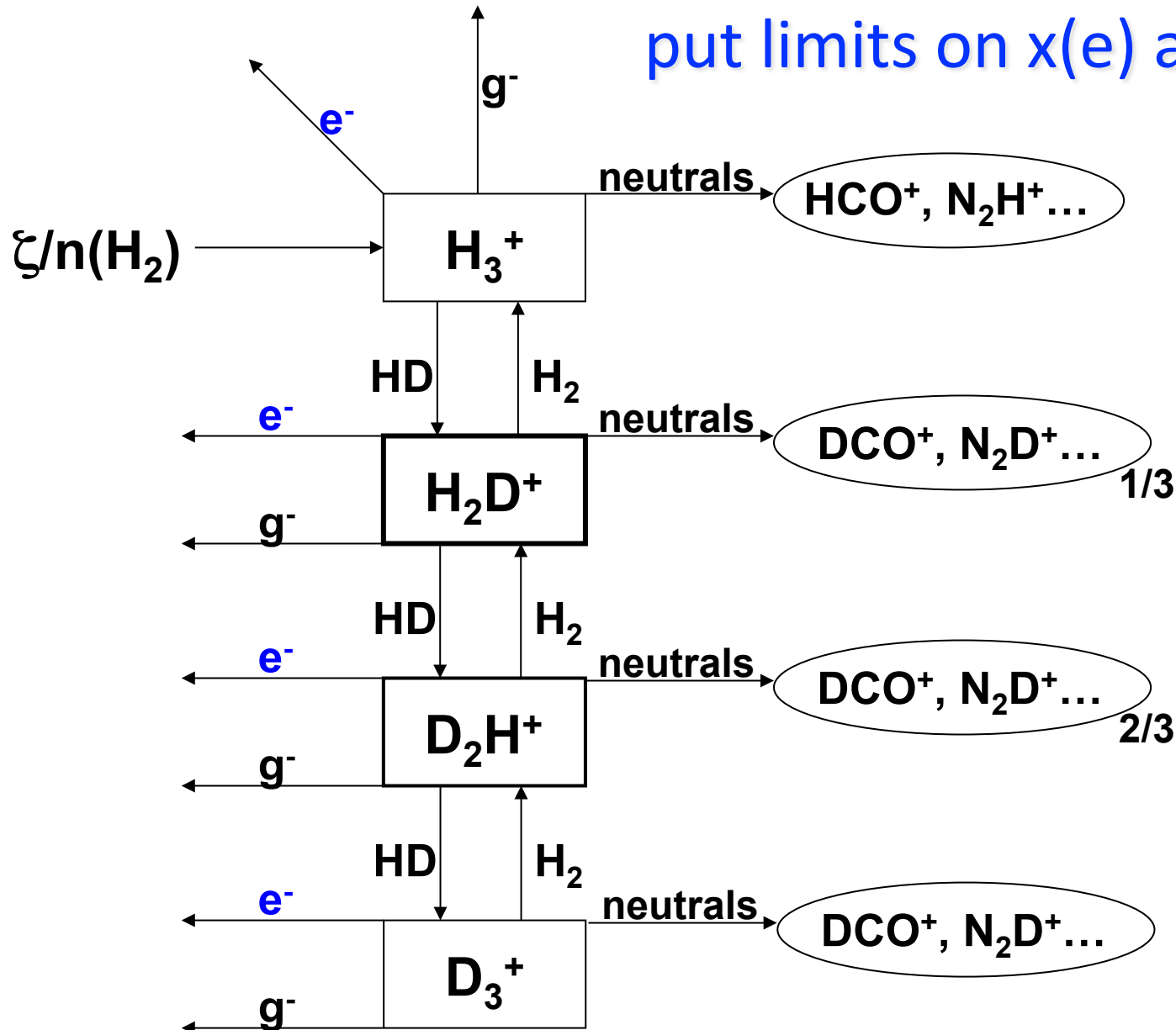
$\text{H}_2\text{D}^+ / \text{H}_3^+$ increases if the abundance of gas phase neutral species decreases (Dalgarno & Lepp 1984):



15

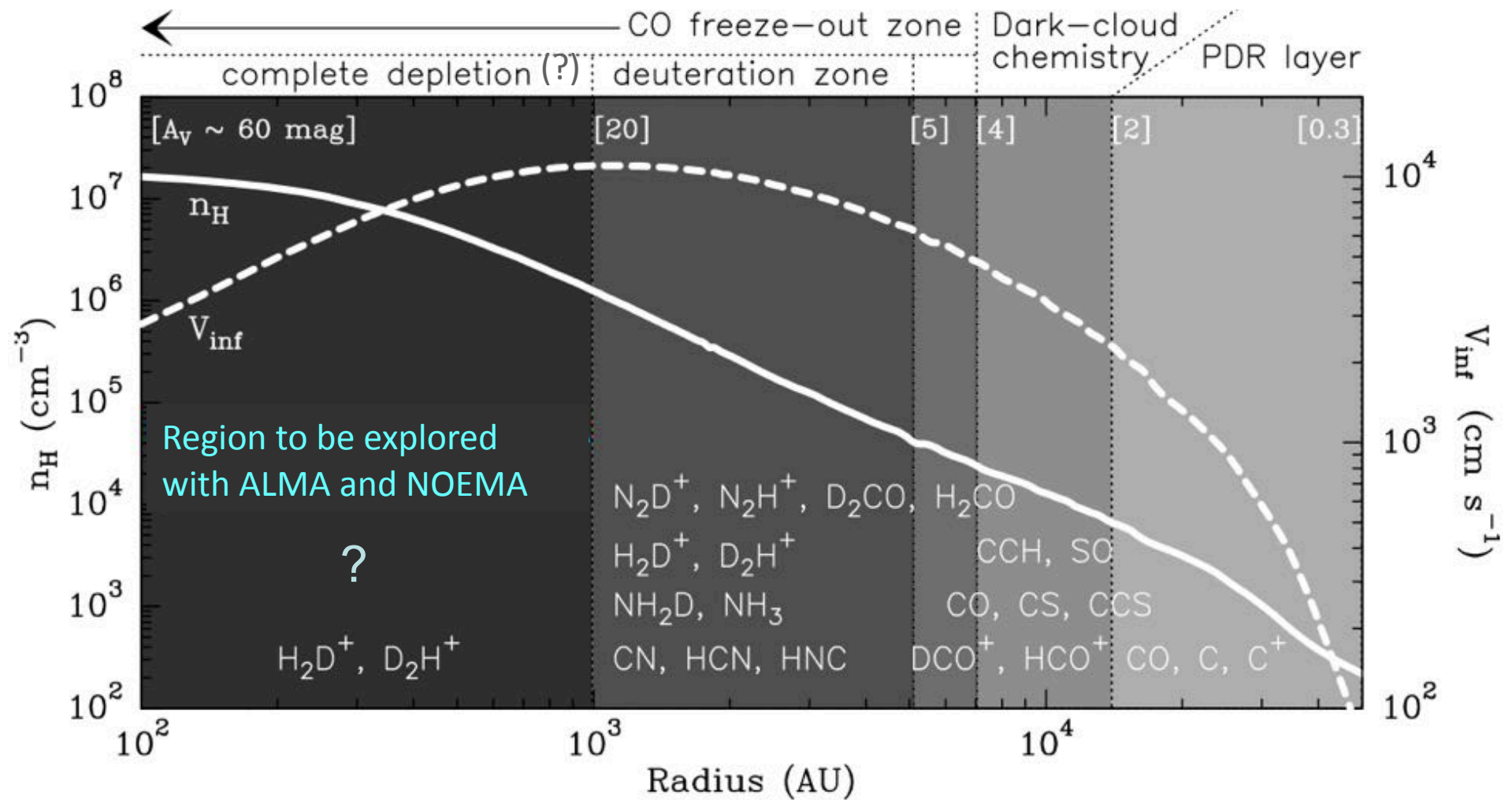


Deuterated molecular ions put limits on $x(e)$ and ζ

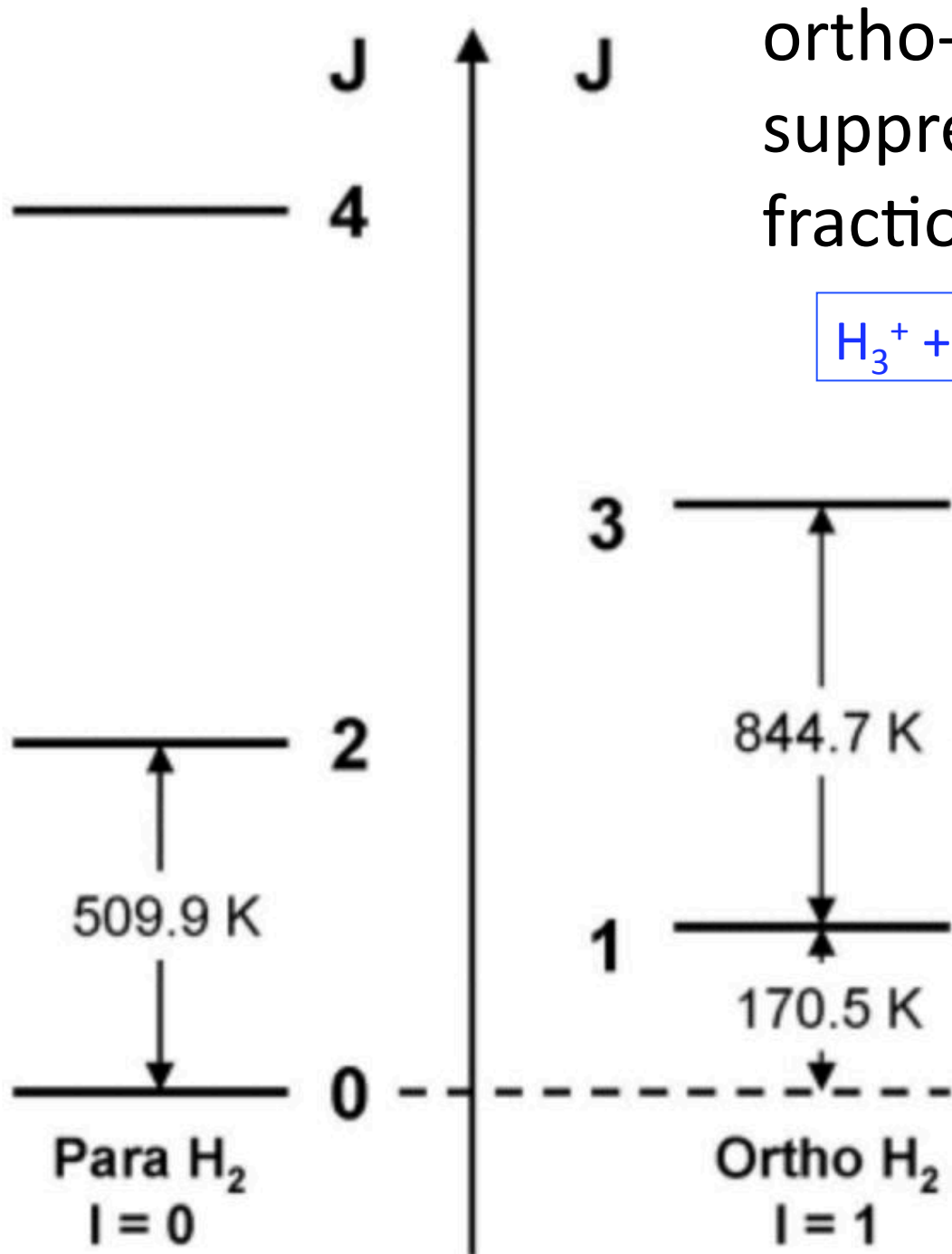


Guelin et al. 1977
Wootten et al. 1979
Guelin et al. 1982
Bergin et al. 1998
Caselli et al. 1998
Dalgarno 2006

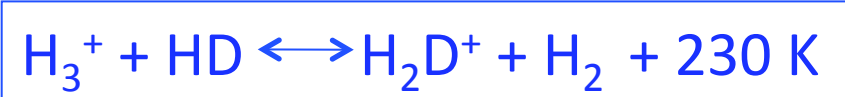
Deuterated molecules are the best tracers of pre-stellar cores



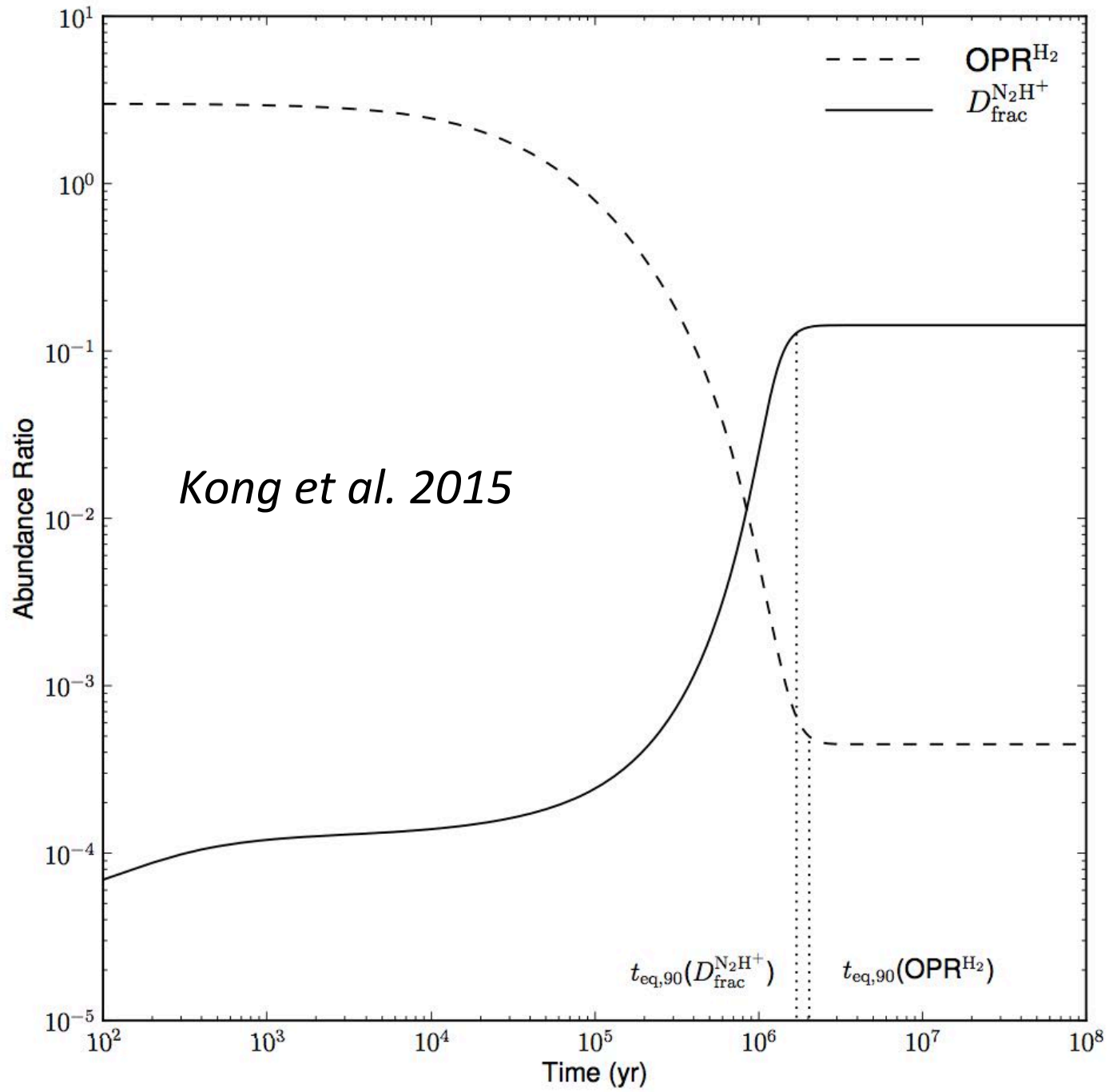
18



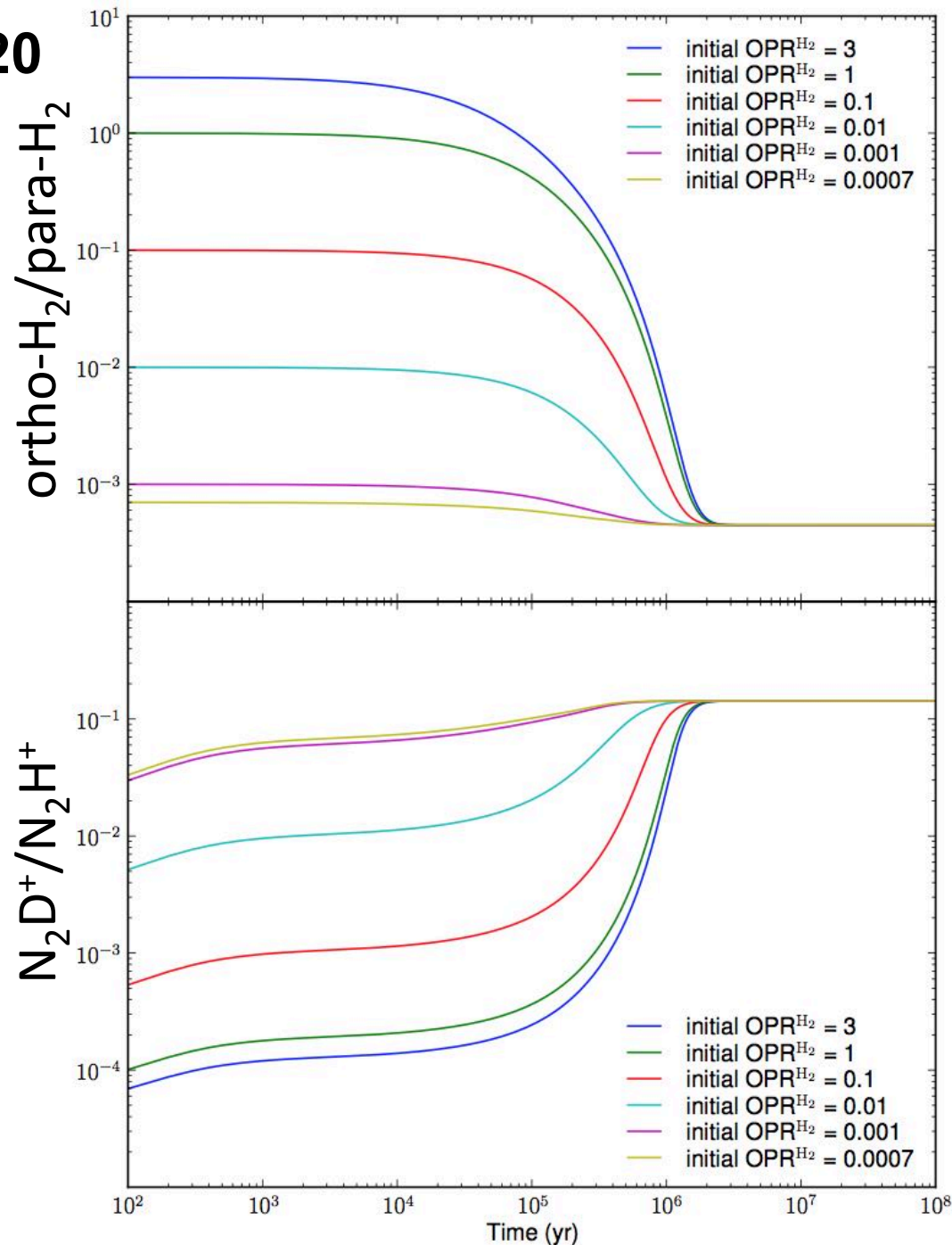
ortho-H₂ can slow down /
suppress the deuterium
fractionation



Pagani et al. 1992
Gerlich et al. 2002
Hugo et al. 2009
Kong et al. 2015



20



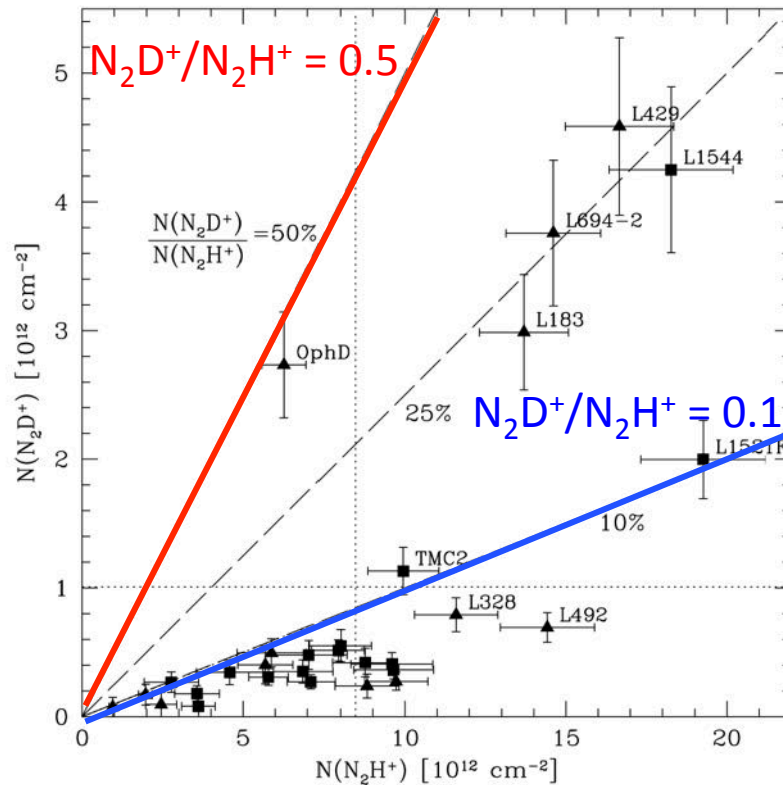
The conversion from ortho to para H_2 is a slow process and it is required to explain the observed large deuterium fractions.

$D_{\text{frac}} > 0.1$, requires collapse to be proceeding at rates about 10 times slower than that of free-fall collapse.

Kong et al. 2015

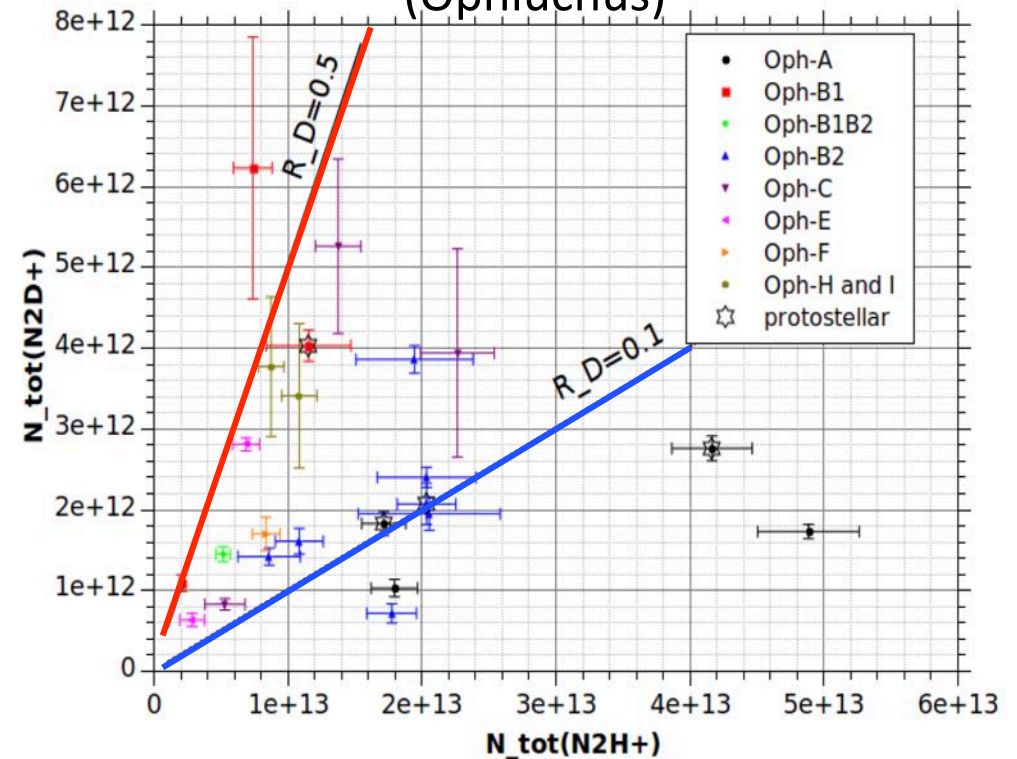
21 Large (variations of) deuterium fractions

Cores in different molecular clouds



Crapsi et al. 2005

Cores in the same molecular cloud
(Ophiuchus)



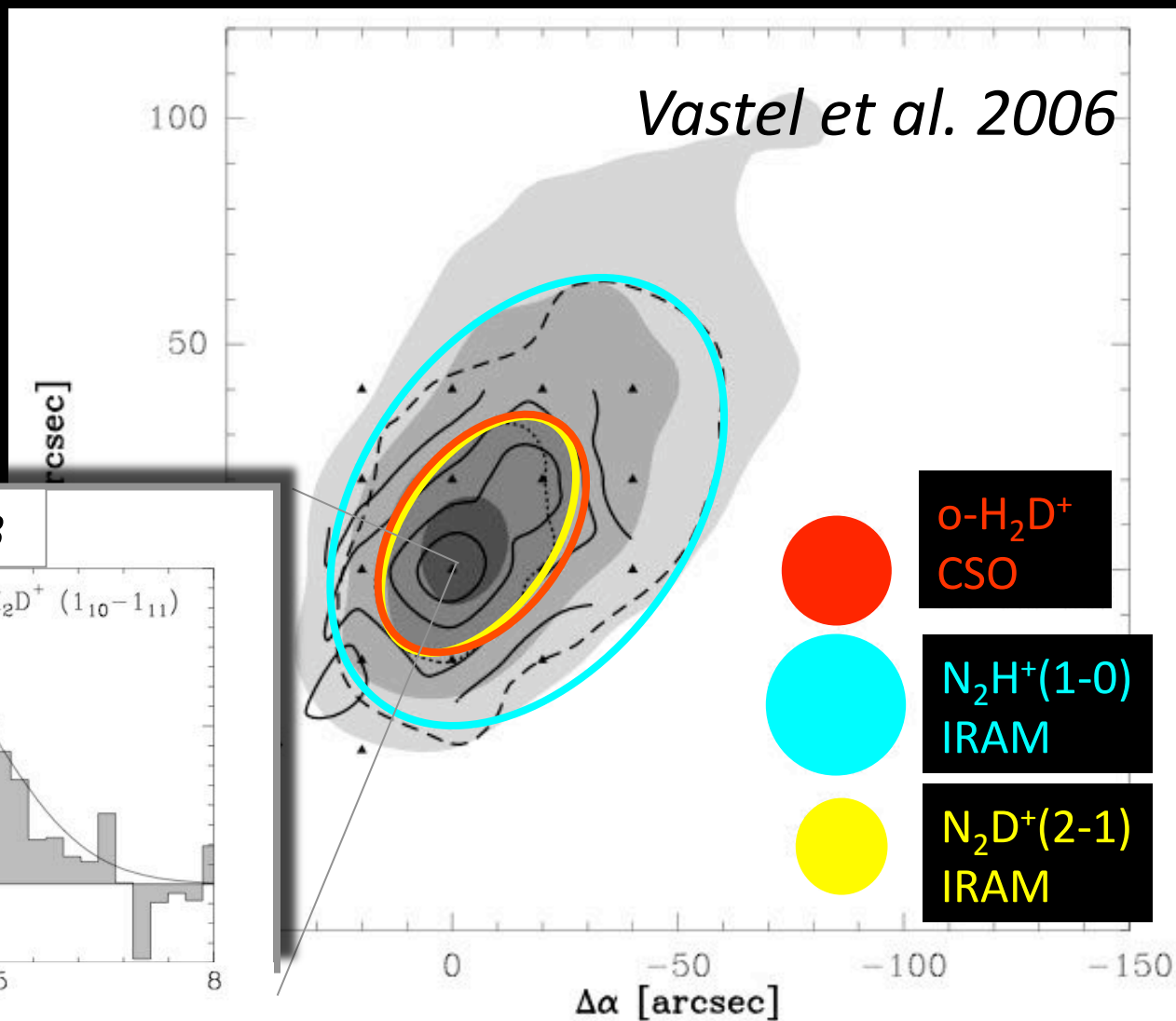
Punanova et al., in prep.

See also Friesen et al. 2010; Friesen, Kirk & Shirley 2013; Schnee et al. 2013

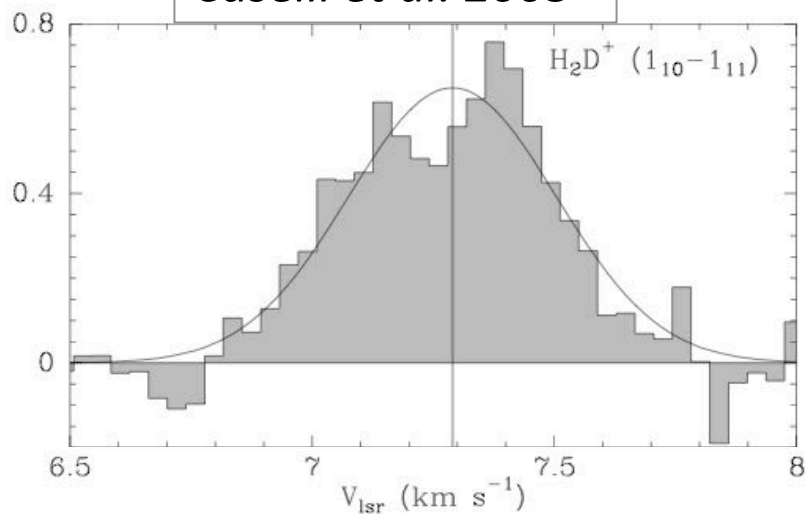
ortho- H_2D^+

The 372 GHz o- H_2D^+ line is strong; its emission is extended ~ 5000 AU

Only models including
all multiply deuterated
forms of H_3^+ can
reproduce these data
(Roberts et al. 2003;
Walmsley et al. 2004;
Aikawa et al. 2005)

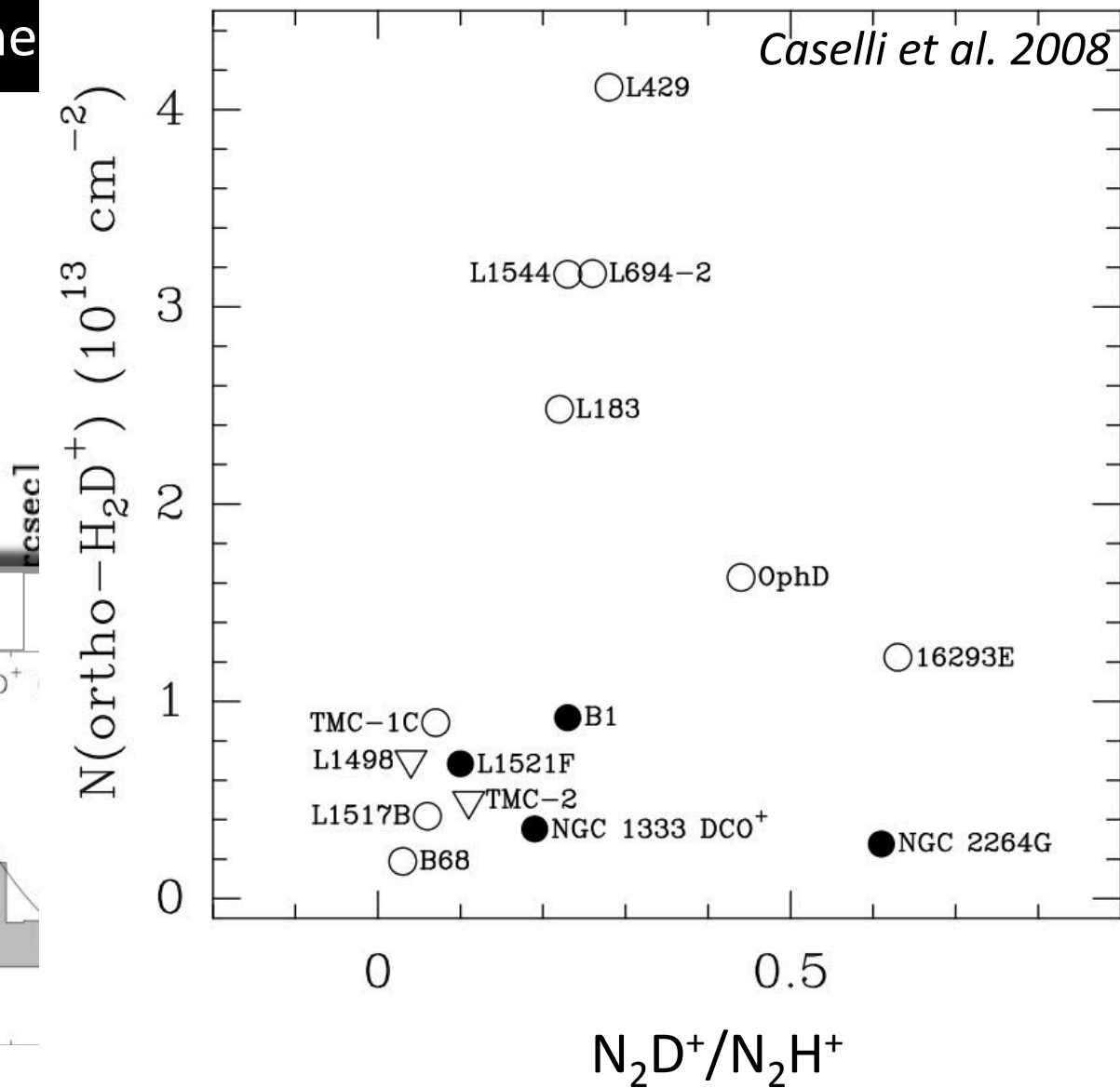
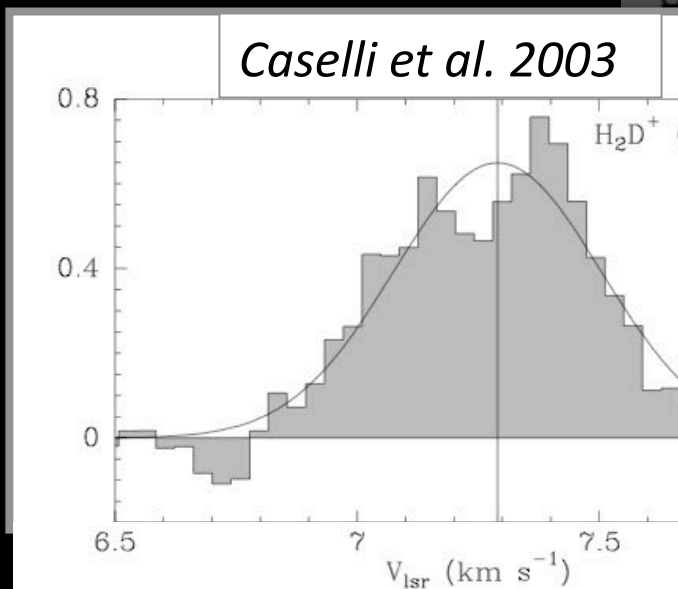


Caselli et al. 2003



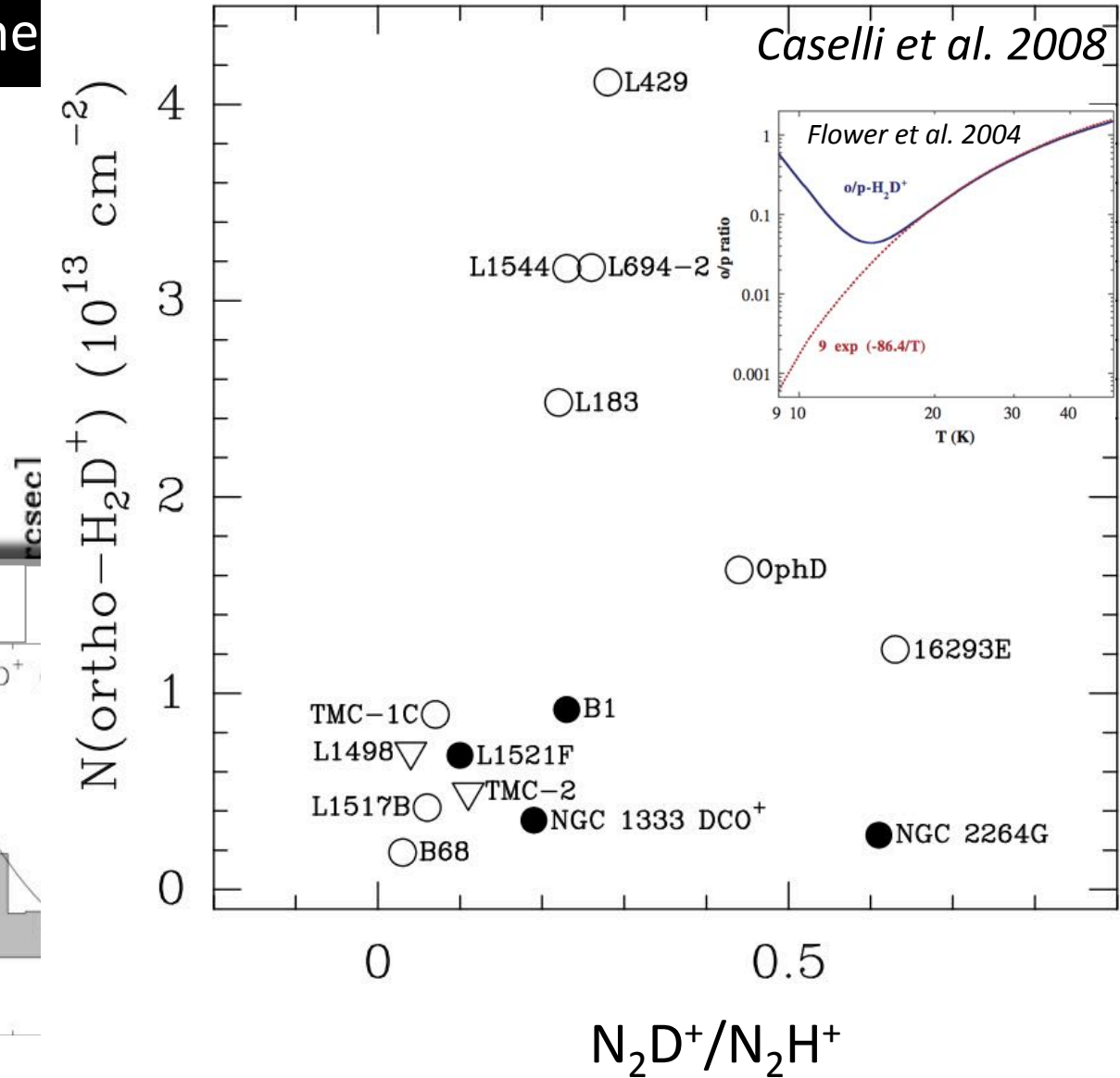
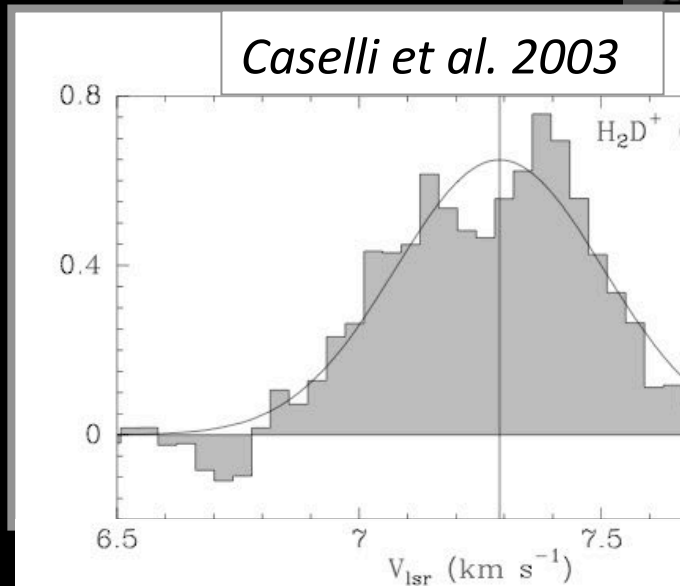
The 372 GHz o- H_2D^+ line

Only models including all multiply deuterated forms of H_3^+ can reproduce these data (Roberts et al. 2003; Walmsley et al. 2004; Aikawa et al. 2005)



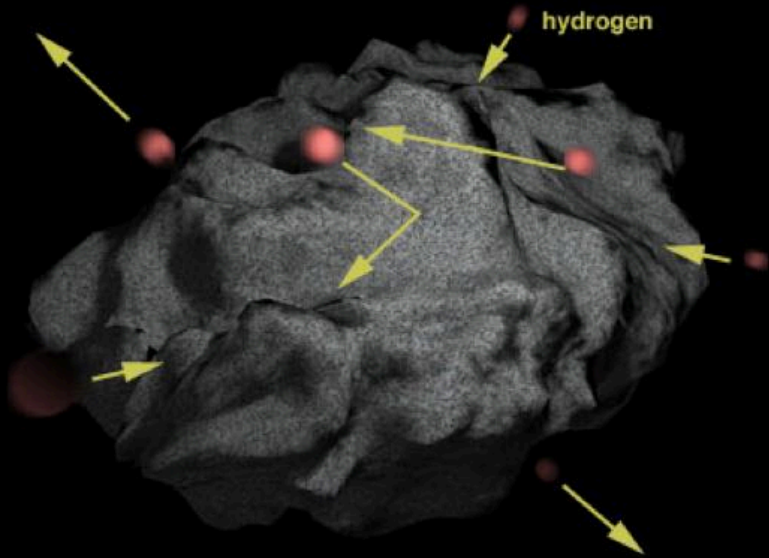
The 372 GHz o- H_2D^+ line

Only models including all multiply deuterated forms of H_3^+ can reproduce these data (Roberts et al. 2003; Walmsley et al. 2004; Aikawa et al. 2005)



25

How much ortho-H₂ is in molecular clouds?



Upon formation: $oH_2/pH_2 = 3$

In diffuse clouds: $oH_2/pH_2 \sim 0.3-0.8$
(Crabtree et al. 2011)

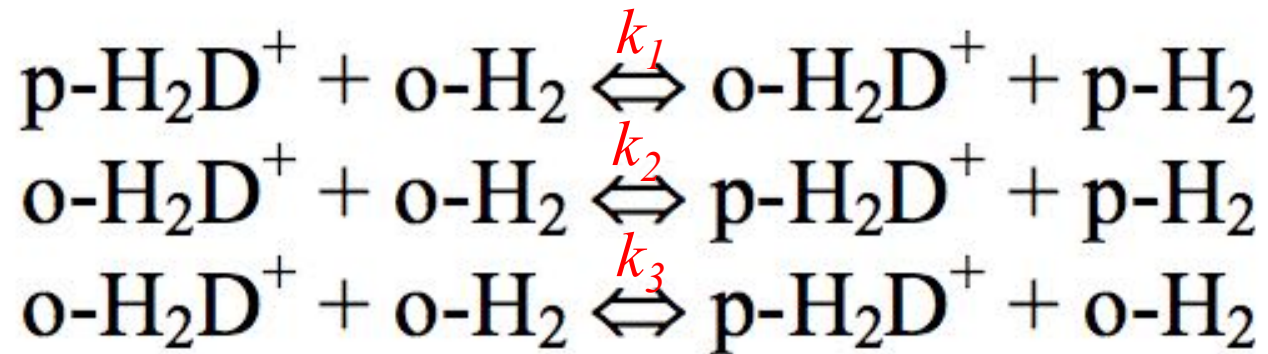
In the pre-stellar core L183:
 $oH_2/pH_2 \sim 0.1$ (Pagani et al. 2009)

In the starless core B68: $oH_2/pH_2 \sim 0.015$
(Maret & Bergin 2007)

In the pre-stellar core L1544: $oH_2/pH_2 \sim 0.003$
(Kong et al. 2015)

26

Analytical relation between the
H₂ and H₂D⁺ ortho-to-para ratios

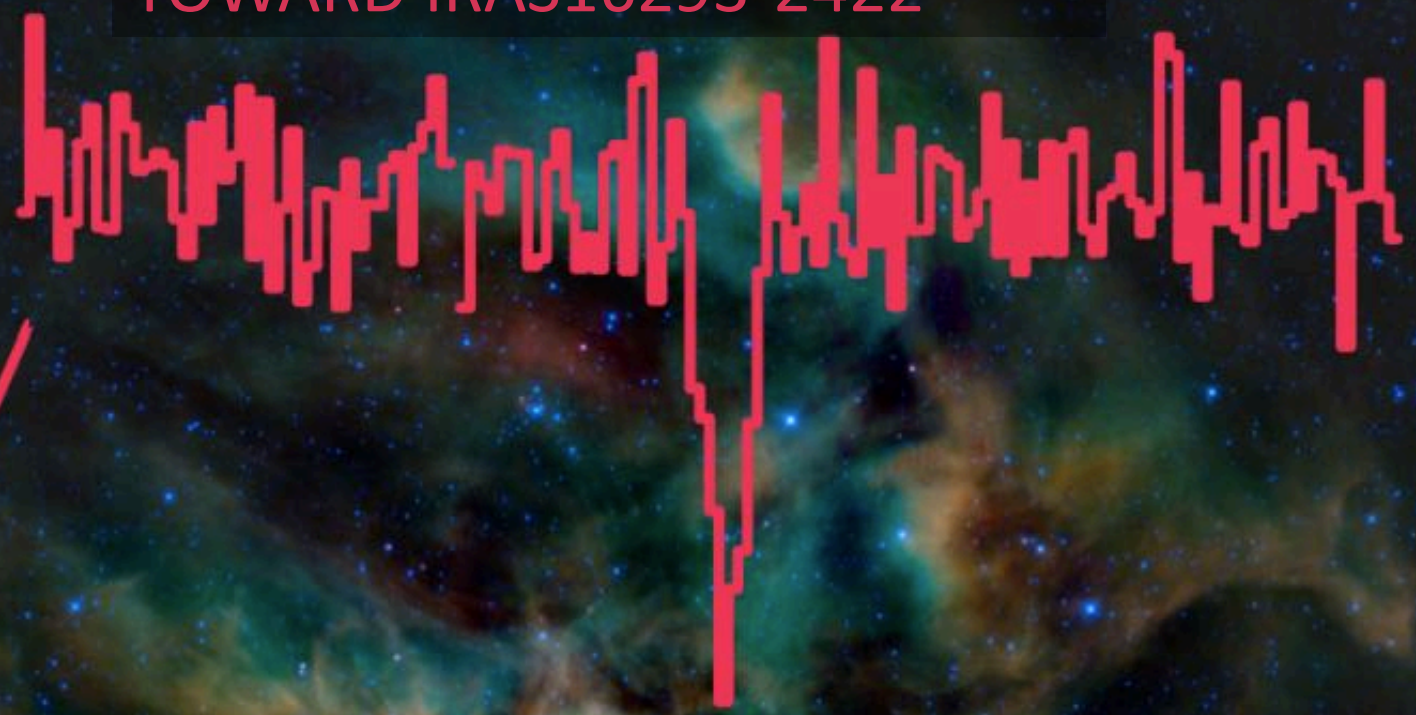


$$\frac{[\text{o-H}_2\text{D}^+]}{[\text{p-H}_2\text{D}^+]} = \frac{(k_1^+ + k_3^-) \times [\text{o-H}_2] / [\text{p-H}_2] + k_2^-}{(k_2^+ + k_3^+) \times [\text{o-H}_2] / [\text{p-H}_2] + k_1^-}$$

Hugo et al. 2009; Brünken et al. 2014

27

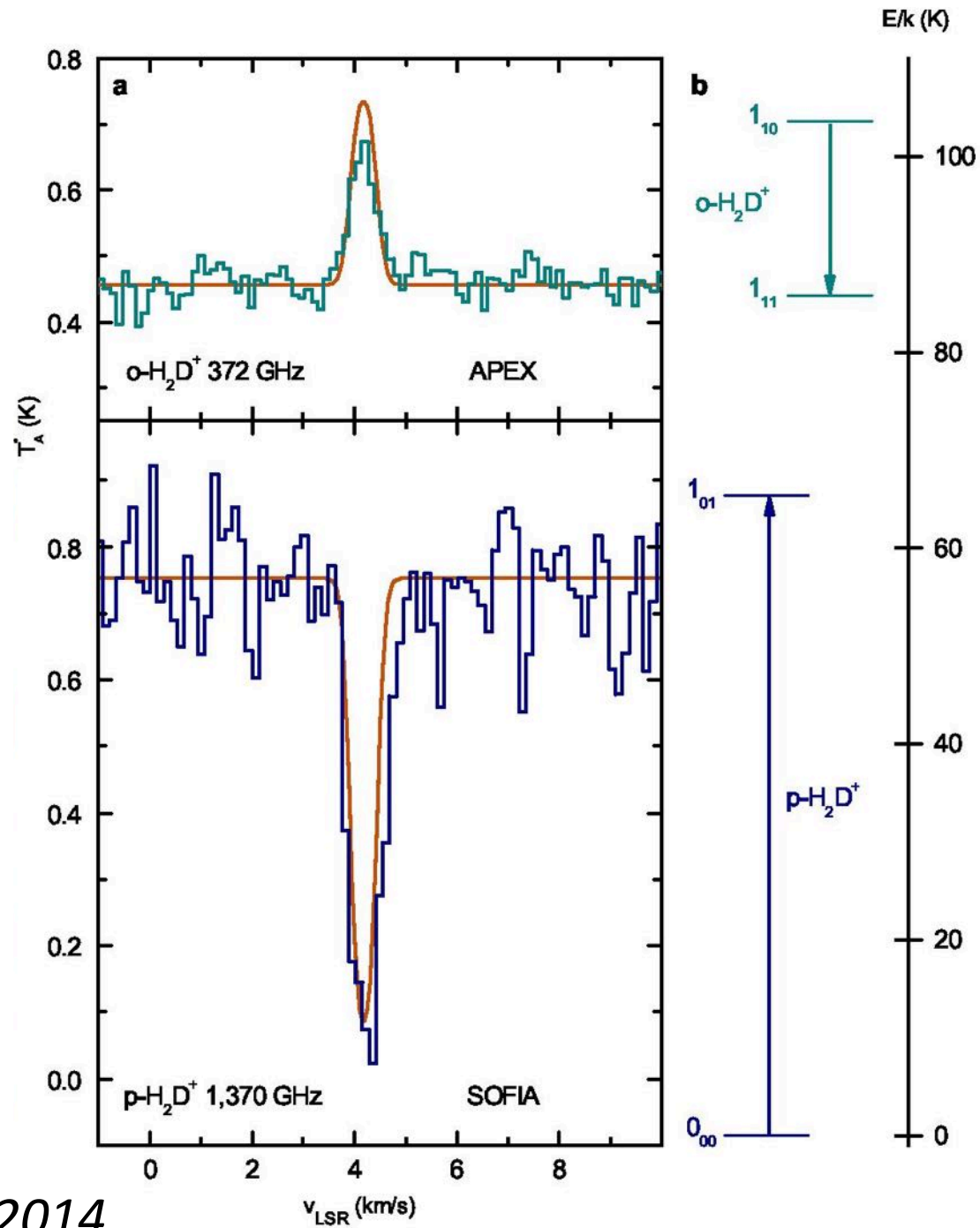
FIRST DETECTION OF para-H₂D⁺
TOWARD IRAS16293-2422

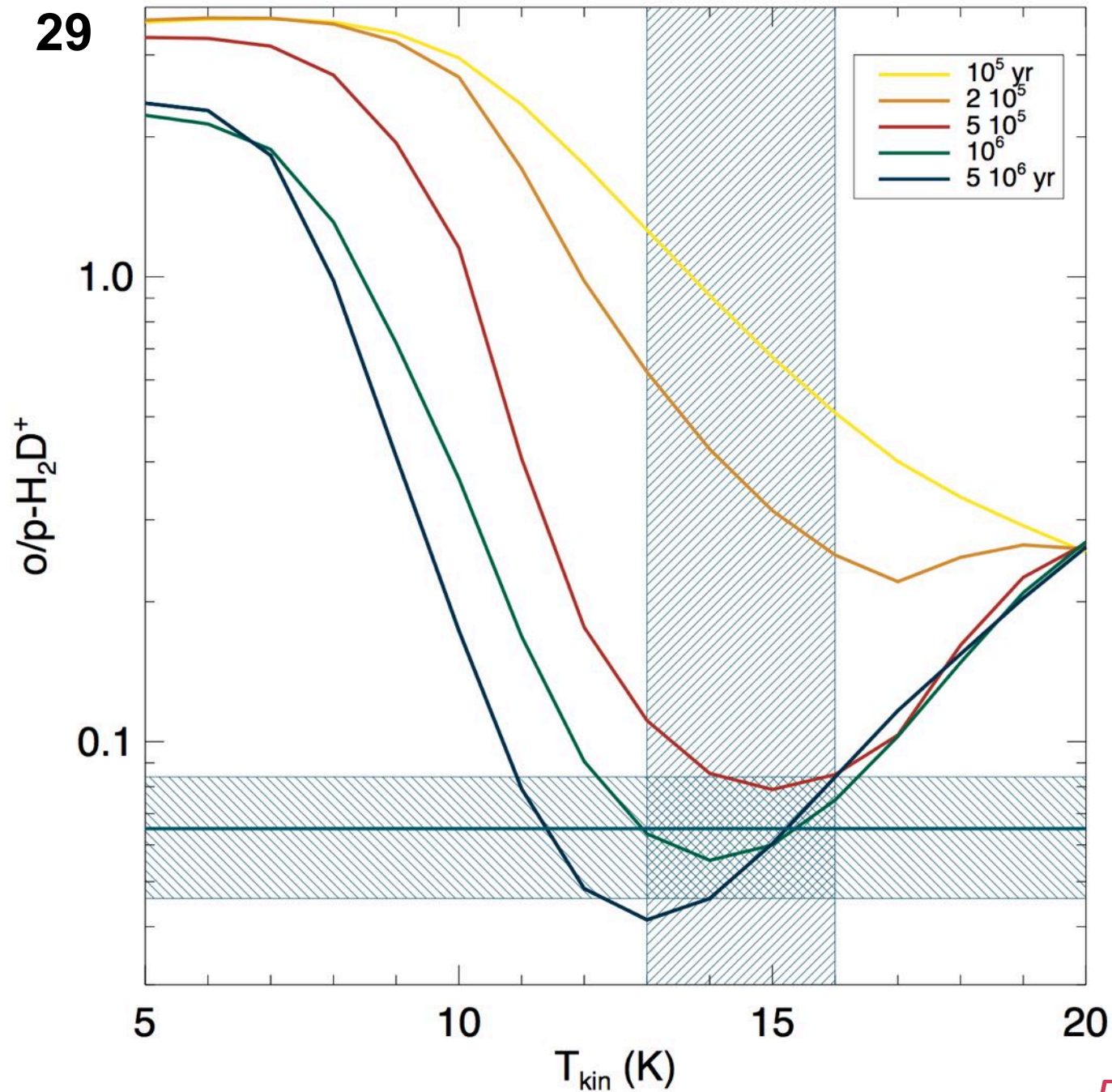


para-H₂D⁺



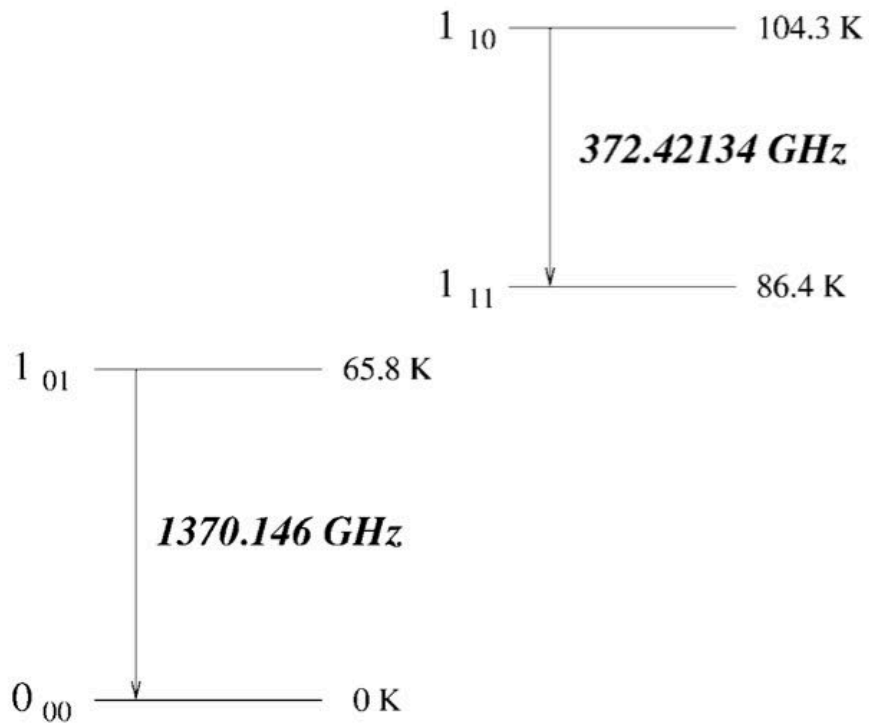
Brünken, Sipilä, Chambers, Harju, Caselli, Asvany, Honingh, Kamiński, Menten, Stutzki, Schlemmer 2014, Nature



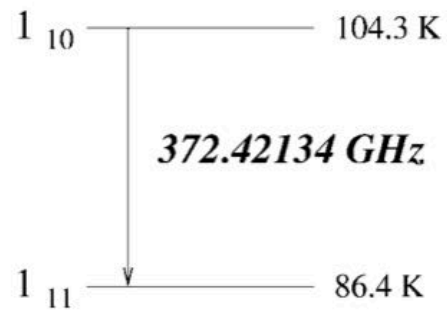


The molecular gas in the cool envelope has been subject to chemical processing for at least one million years.

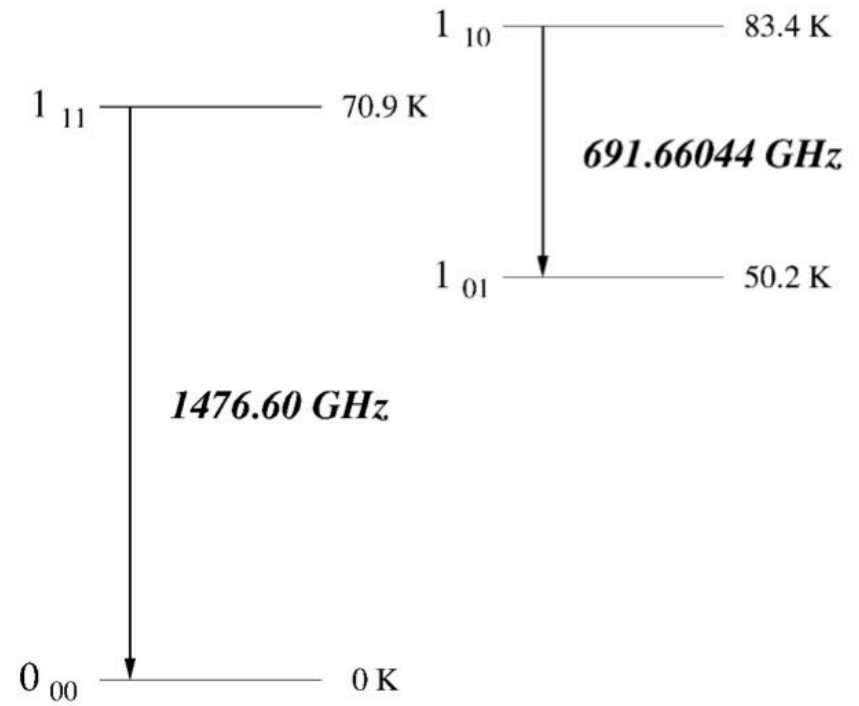
30



PARA H_2D^+



ORTHO H_2D^+



ORTHO D_2H^+

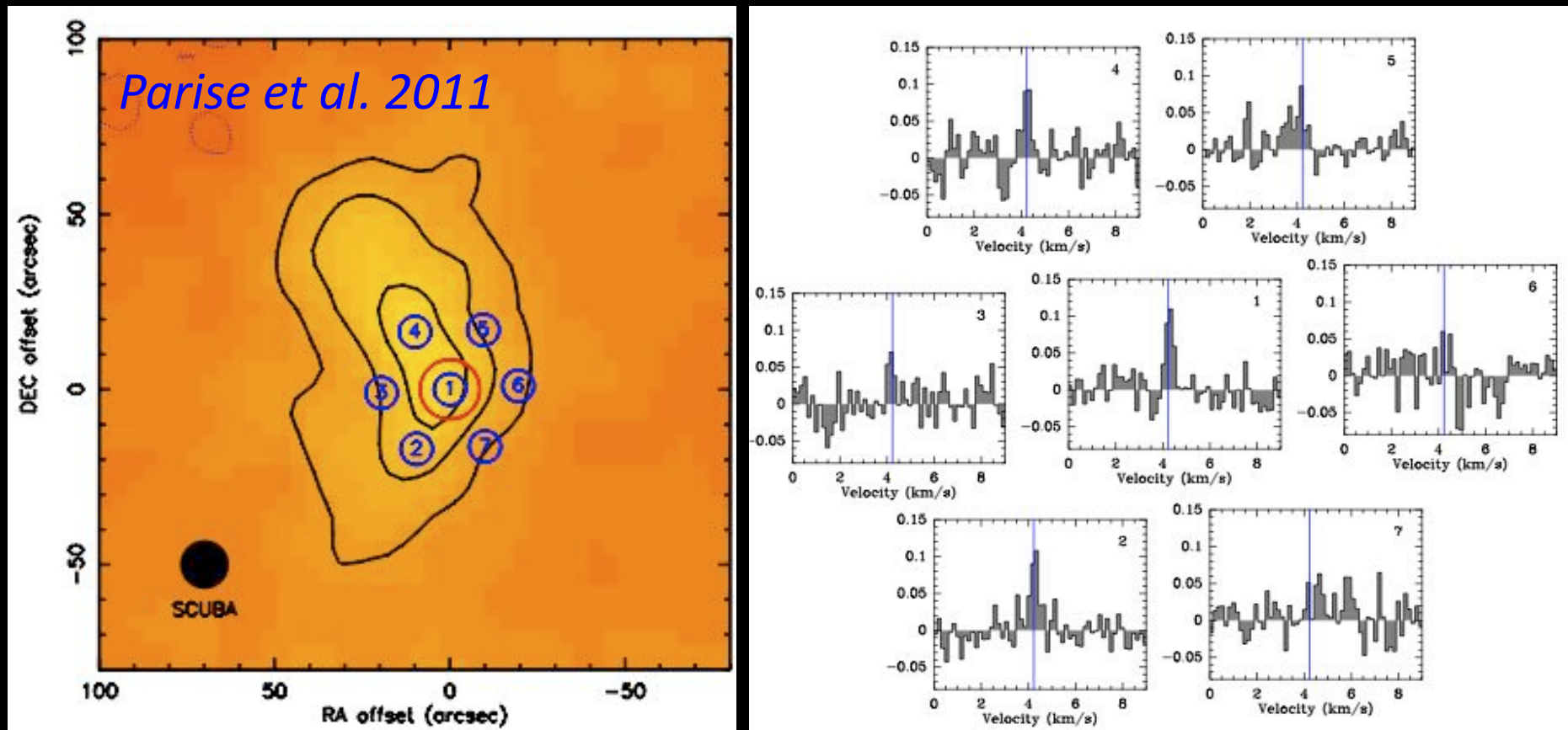


PARA D_2H^+

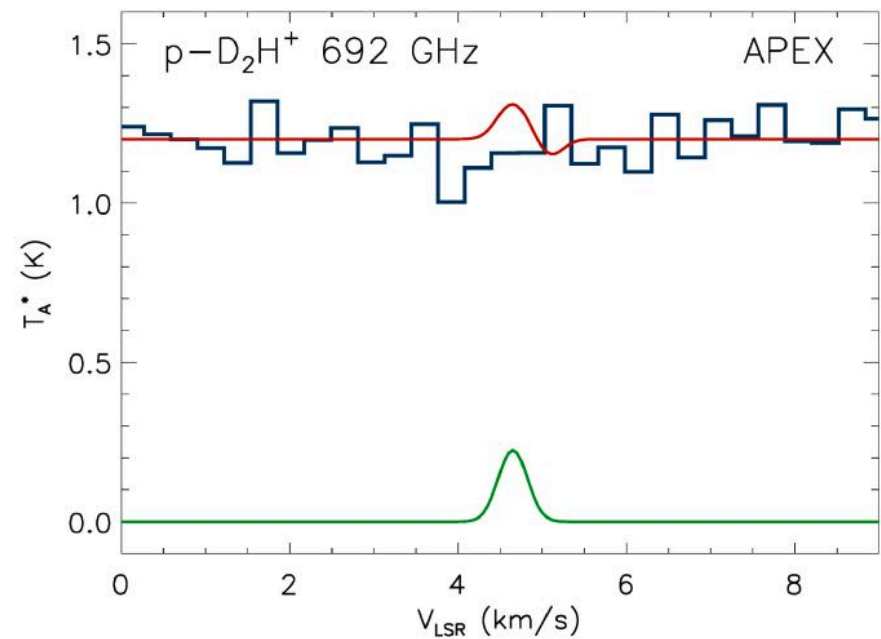
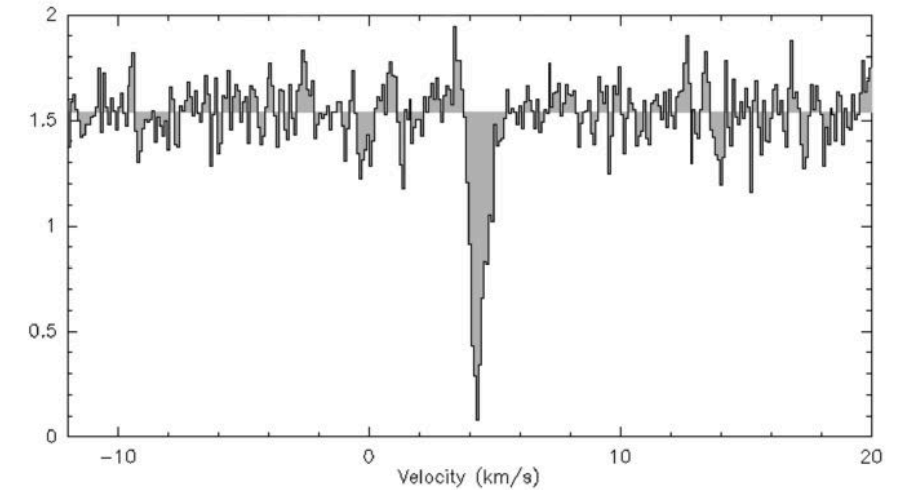
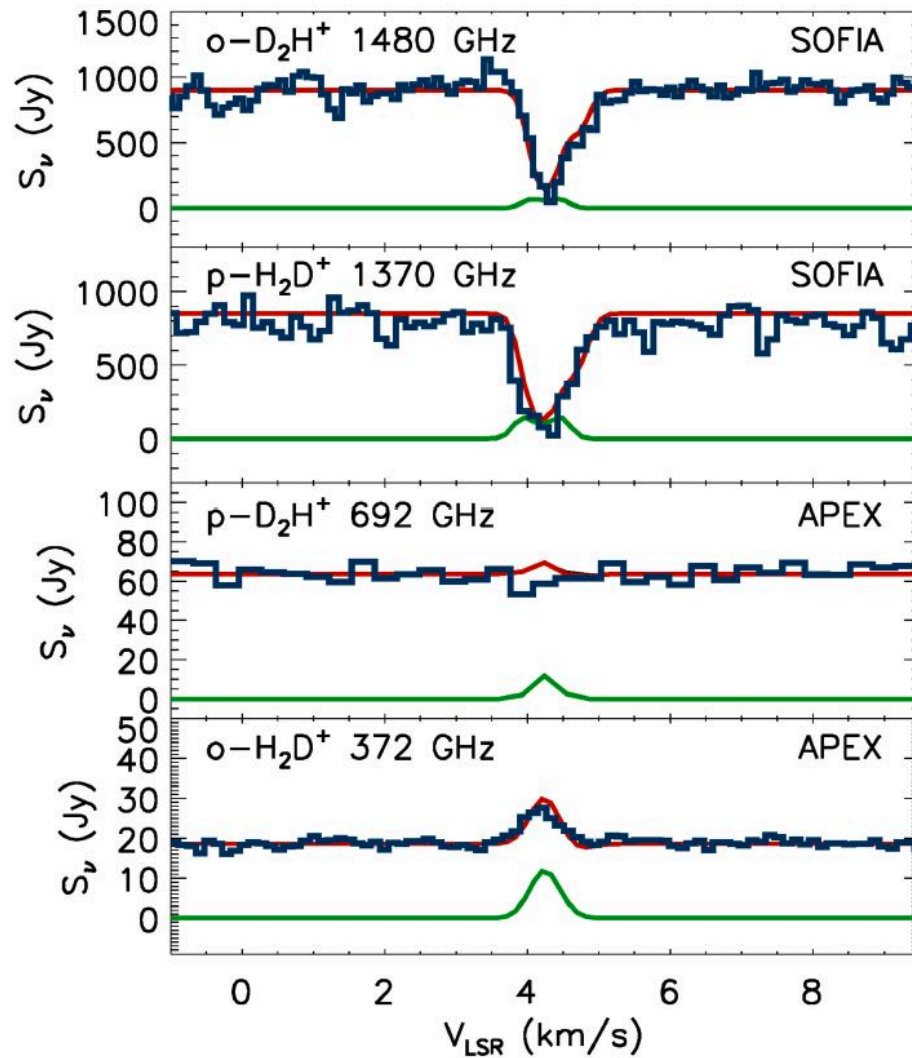


from Vastel et al. 2004

Extended para-D₂H⁺ emission ($\sim 40'' \sim 5000$ AU) toward L1688/H-MM1 (692 GHz; APEX-CHAMP+)



See Vastel et al. (2004) for first detection of para-D₂H⁺

ortho-D₂H⁺ with SOFIA

Ro-vibrational lines

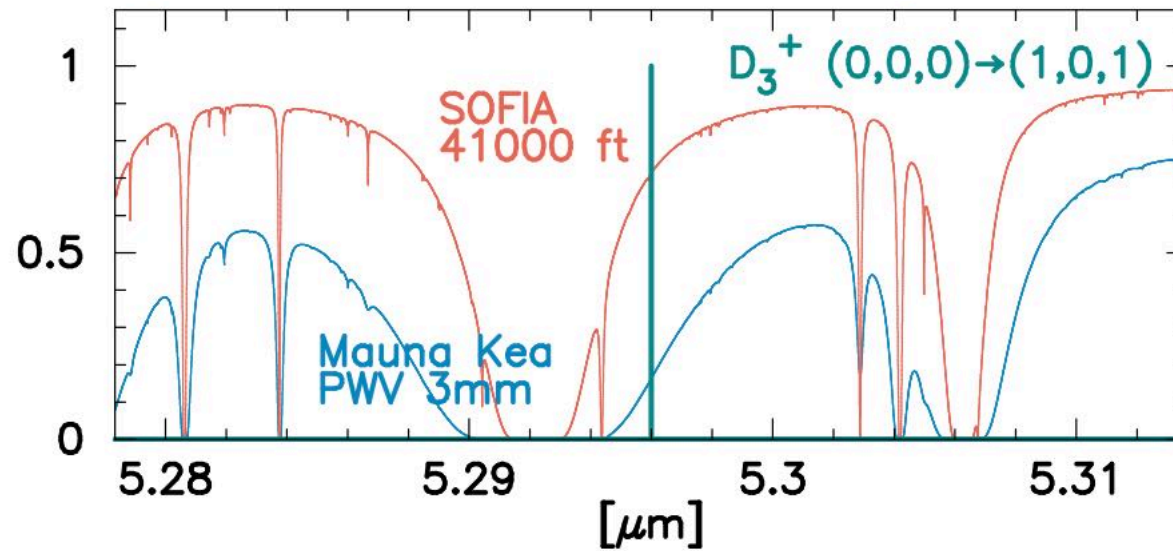
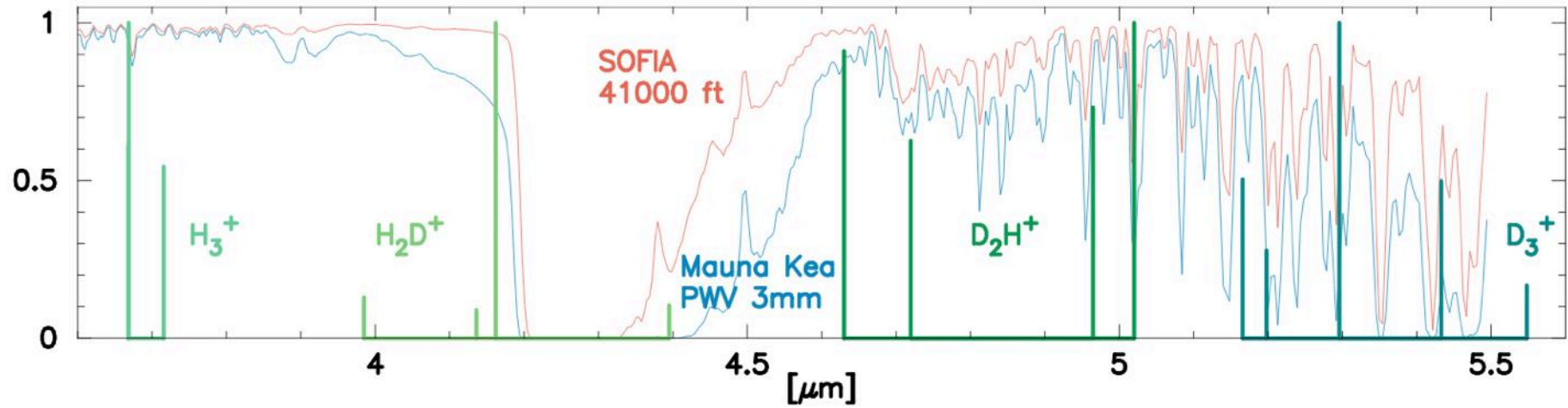
Table 3. Requisite column densities for unit optical depth in the absorption line centre for the deuterated forms of H_3^+ . Spectroscopic data from Ramanlal & Tennyson (2004).

Species	Upper level	Lower level	E_l (cm^{-1})	Wavelength (μm)	$N_{\tau=1}$ (cm^{-2})
H_2D^+ (para)	(1, 1, 1) ^a	(0, 0, 0)	0.0	4.162	1.9×10^{14}
H_2D^+ (ortho)	(0, 0, 0)	(1, 1, 1)	60.0	4.395	6.1×10^{14}
H_2D^+ (ortho)	(2, 0, 2)	(1, 1, 1)	60.0	4.136	7.1×10^{14}
H_2D^+ (ortho)	(2, 2, 0)	(1, 1, 1)	60.0	3.985	4.9×10^{14}
D_2H^+ (ortho)	(1, 1, 1)	(0, 0, 0)	0.0	4.965	4.7×10^{14}
D_2H^+ (ortho)	(1, 0, 1)	(0, 0, 0)	0.0	4.72	5.5×10^{14}
D_2H^+ (para)	(1, 1, 0)	(1, 0, 1)	34.9	5.02	6.9×10^{14}
D_2H^+ (para)	(2, 0, 2)	(1, 0, 1)	34.9	4.63	7.6×10^{14}
D_3^+ (ortho)	(0, 1, 1, 0, 1) ^b	(0, 0, 0, 0, 0)	0.0	5.296	2.2×10^{13}
D_3^+ (meta)	(0, 1, 0, 1, 1)	(0, 0, 1, 1, 0)	32.3	5.548	1.6×10^{14}
D_3^+ (meta)	(0, 1, 2, 1, -1)	(0, 0, 1, 1, 0)	32.3	5.198	9.9×10^{13}
D_3^+ (meta)	(0, 1, 2, 1, 1)	(0, 0, 1, 1, 0)	32.3	5.166	9.1×10^{13}
D_3^+ (para)	(0, 1, 1, 0, -1)	(0, 0, 1, 0, 0)	43.6	5.433	4.5×10^{14}

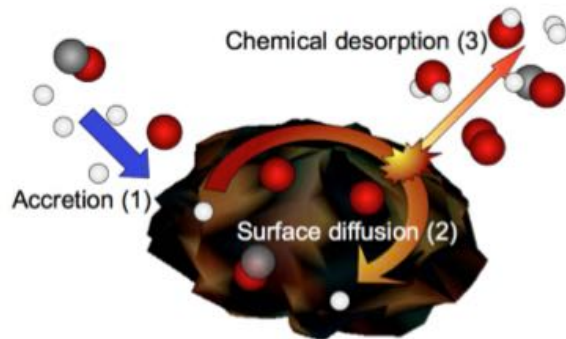
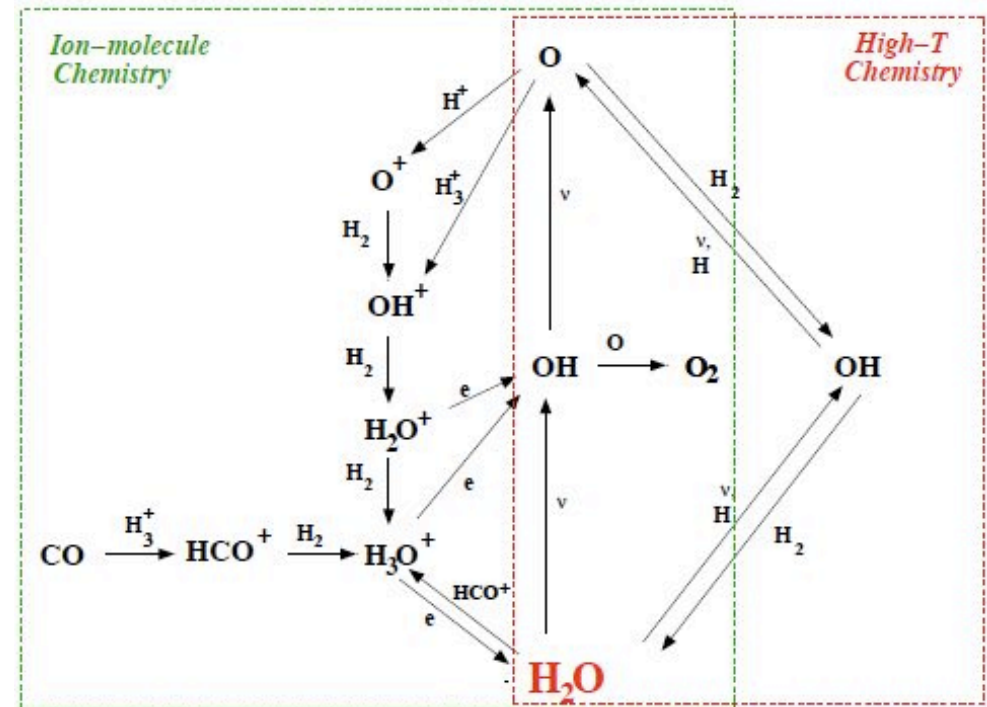
^a (J, K_a, K_c).

^b (ν_1, ν_2, J, G, U).

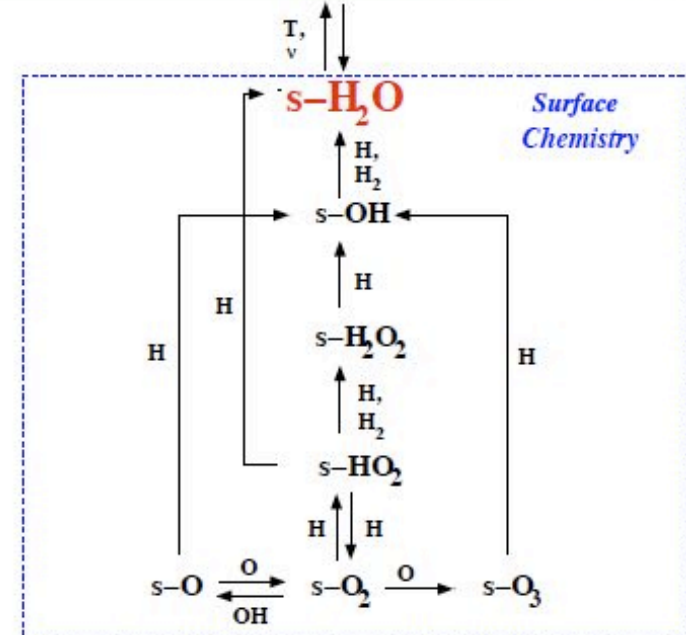
Ro-vibrational lines

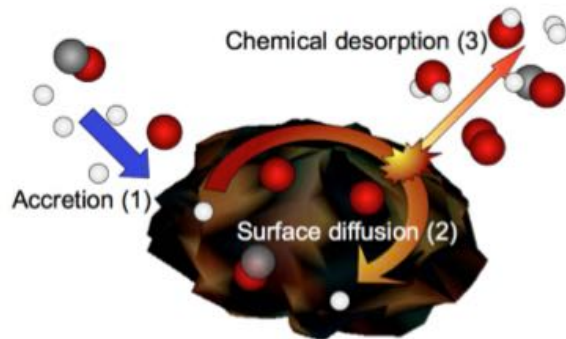
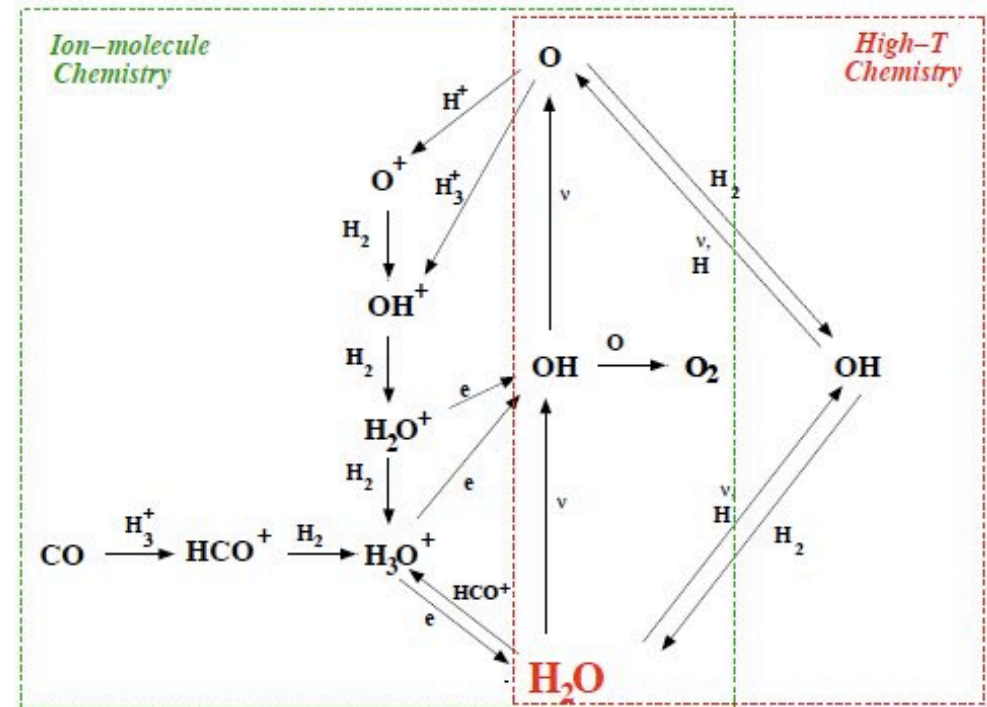
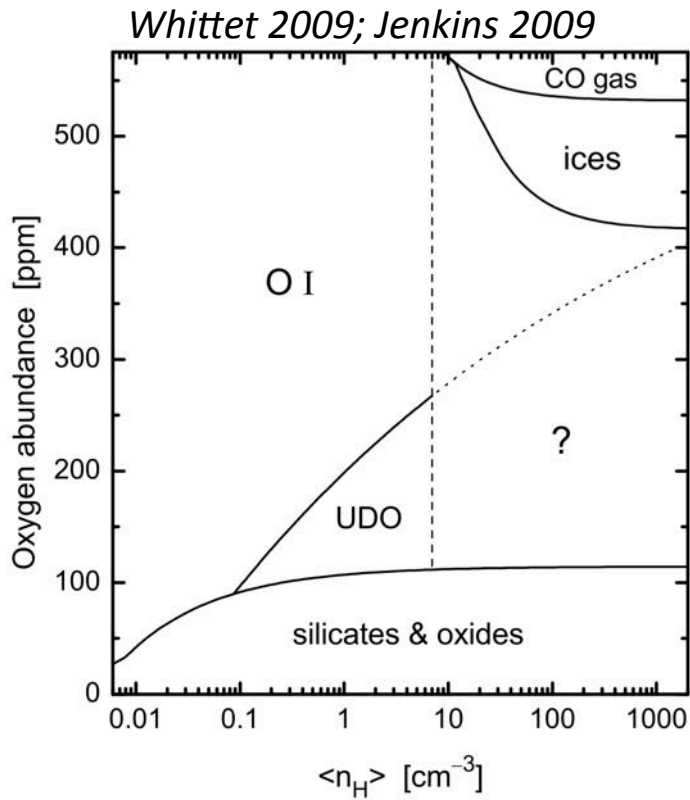


From SOFIA/EXES proposal of Goto et al.

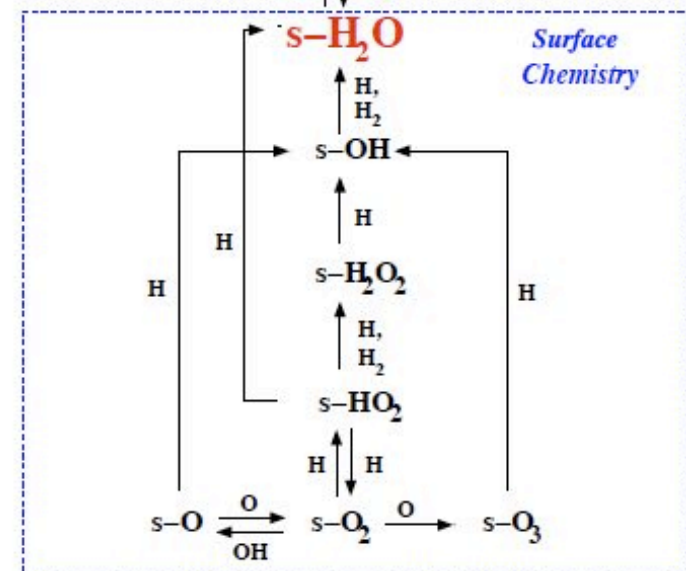


Tielens & Hagen 1982;
 Cuppen & Herbst 2007;
 Miyauchi et al. 2008; Ioppolo
 et al. 2008, 2010; Cazaux et
 al. 2010, 2011; Taquet et al.
 2013; Dulieu et al. 2013

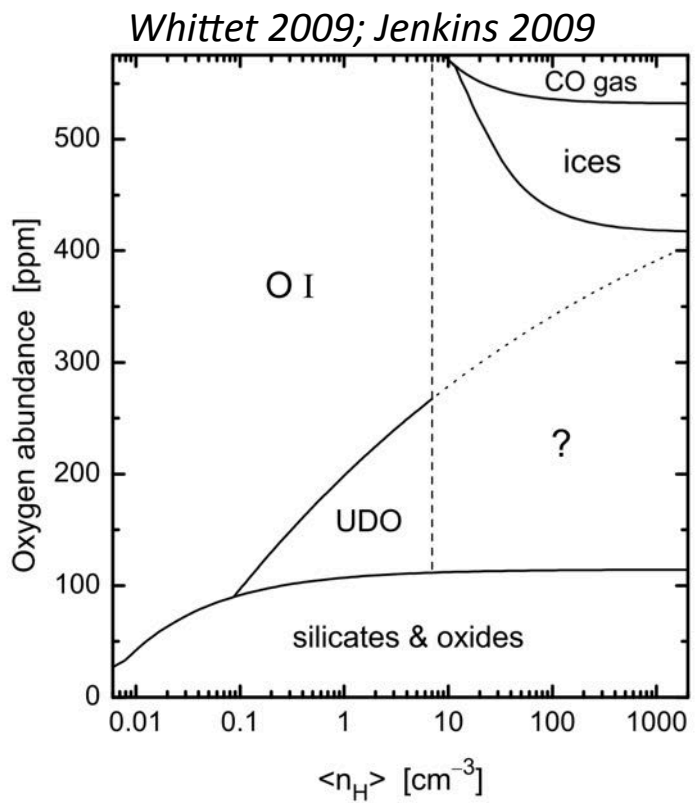




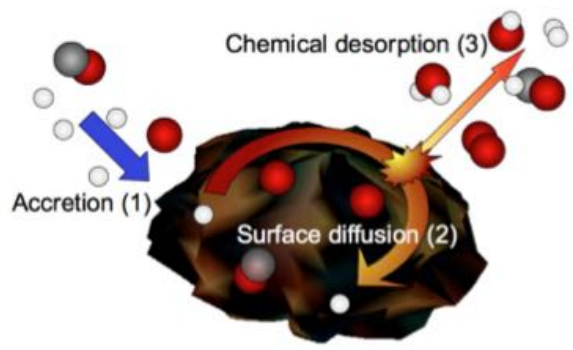
Tielens & Hagen 1982;
 Cuppen & Herbst 2007;
 Miyauchi et al. 2008; Ioppolo
 et al. 2008, 2010; Cazaux et
 al. 2010, 2011; Taquet et al.
 2013; Dulieu et al. 2013



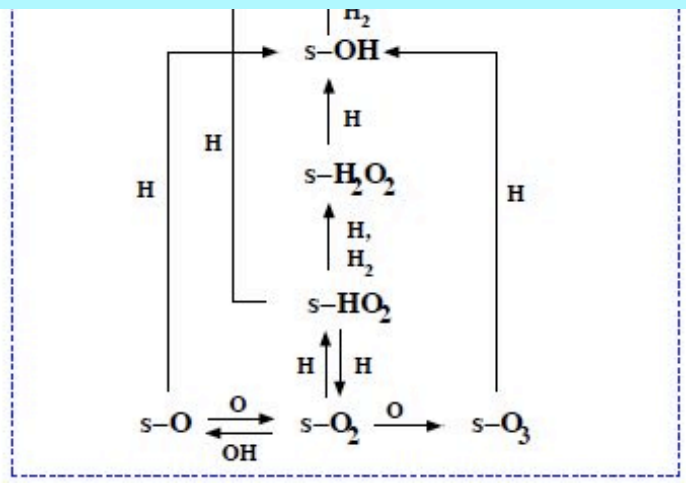
37 Oxygen Chemistry



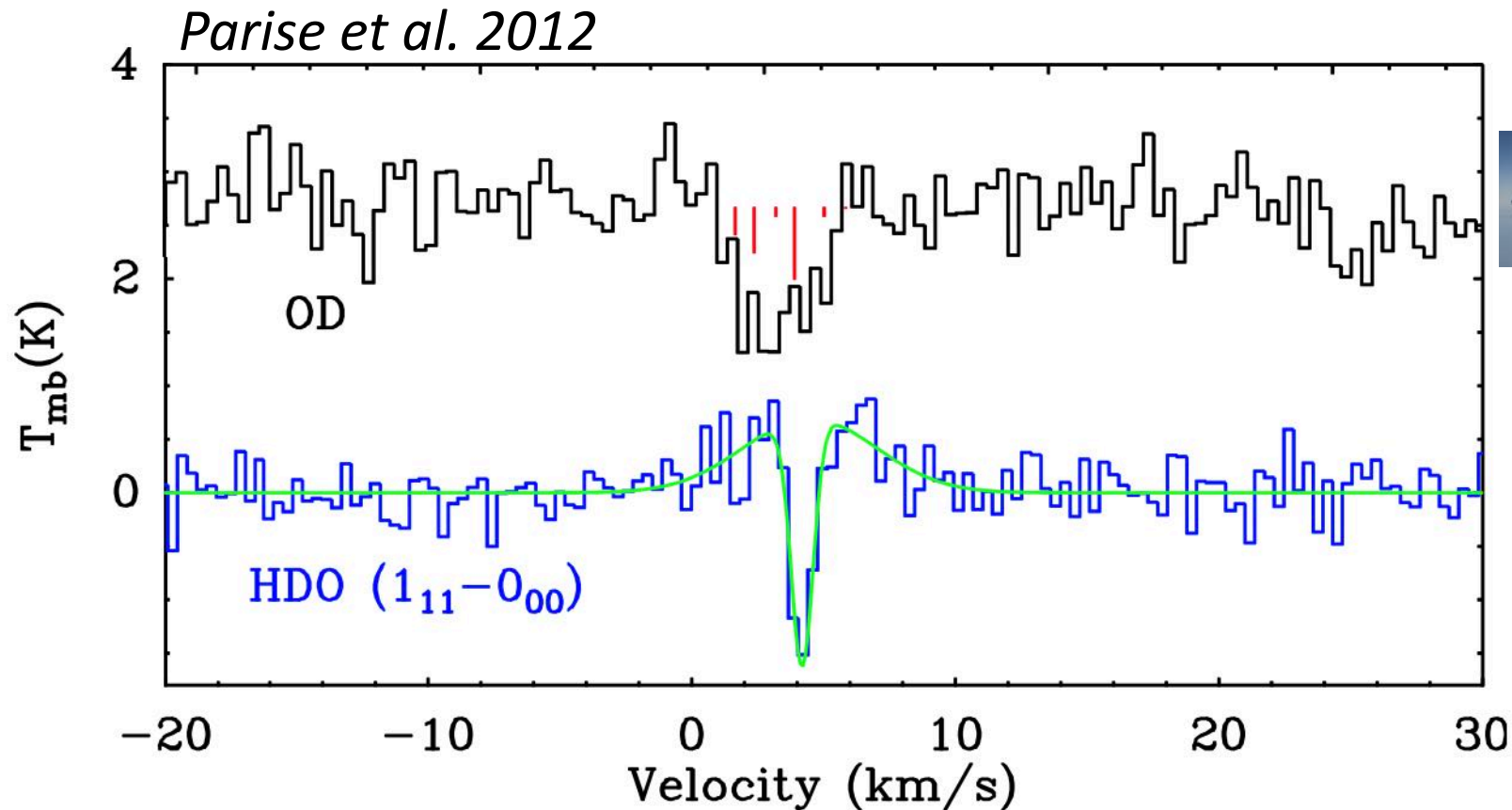
*It will be important in the future to obtain spectra of the highest quality for stars in lines of sight that sample diffuse-ISM dust over a range of visual extinctions, especially in the waveband containing the 5.85 μ m carbonyl feature as it should provide the most sensitive quantitative test for oxygenated organics. Both the **Stratospheric Observatory for Infrared Astronomy** and the **James Webb Space Telescope** will have instruments well matched to this task. – Whittet 2009*



Tielens & Hagen 1982;
Cuppen & Herbst 2007;
Miyachi et al. 2008; Ioppolo et al. 2008, 2010; Cazaux et al. 2010, 2011; Taquet et al. 2013; Dulieu et al. 2013



38 First detection of OD outside the Solar System



OD/HDO between 17 and 90 (high compared to model predictions → gas phase reprocessing through dissociative recombination of H_2DO^+ ?)

Parallel observations of OD and OH could provide valuable constraints on the formation and fractionation of water.

The mercapto radical

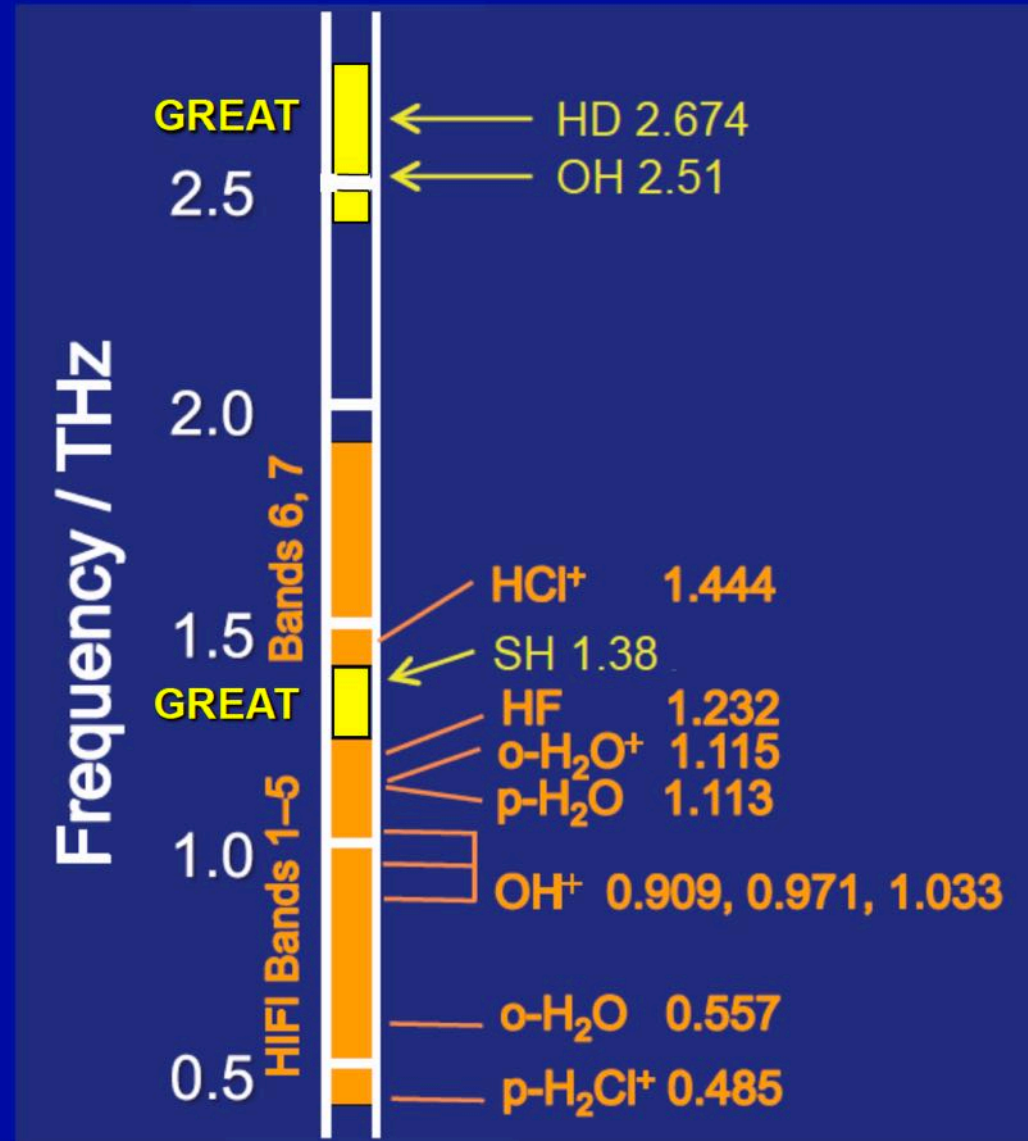
- Cold SH is unobservable from the ground
- The “ground state” rotational transition



falls right in the gap between Bands 5 and 6 of *Herschel*'s HIFI spectrometer

The mercapto radical

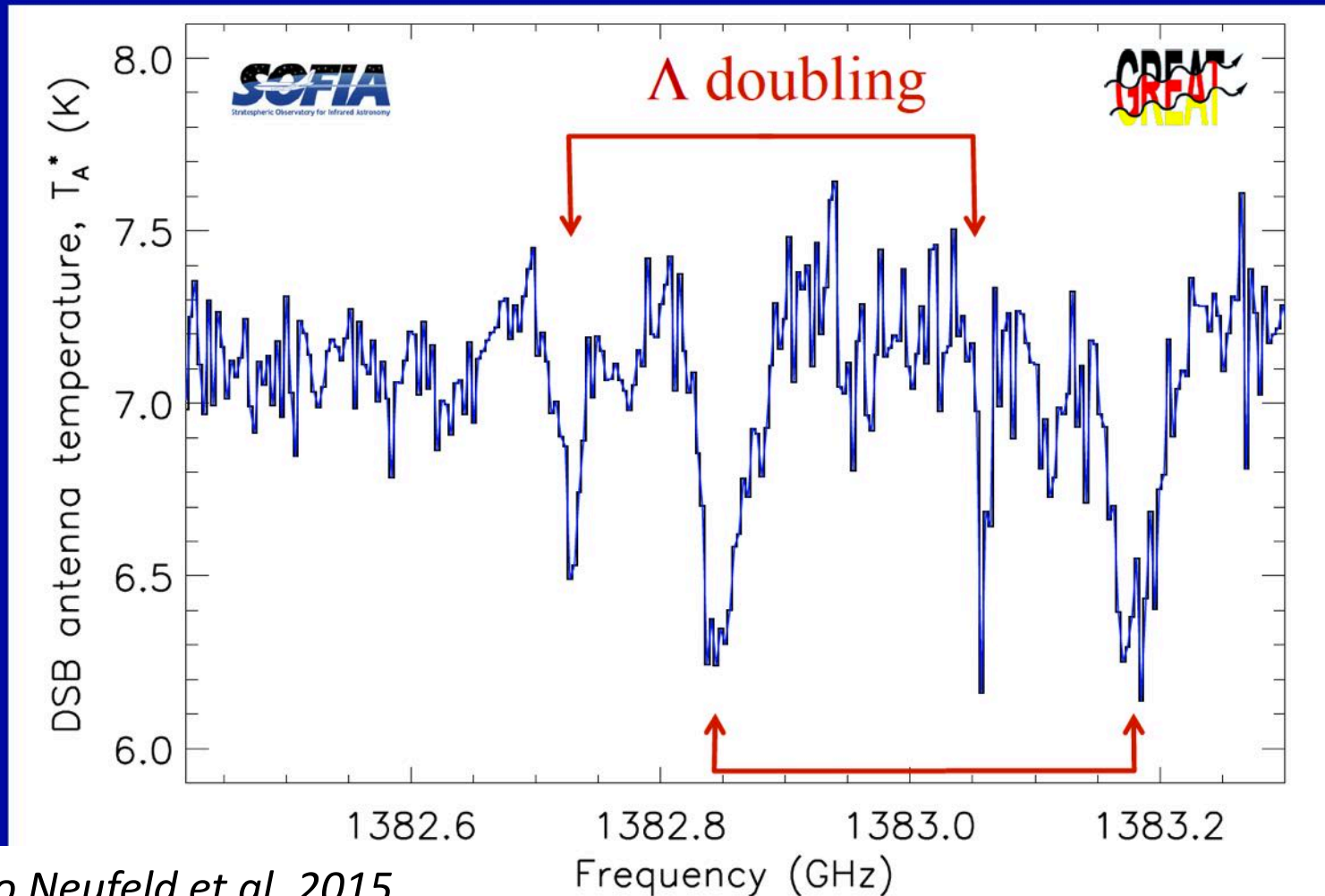
- The GREAT (German Receiver for Astronomy at Terahertz frequencies) spectrometer on SOFIA has a receiver designed to cover this gap in *Herschel*/HIFI coverage (1250 – 1410 GHz)



Courtesy of David Neufeld

Cycle 0: SH clearly detected in absorption toward W49N

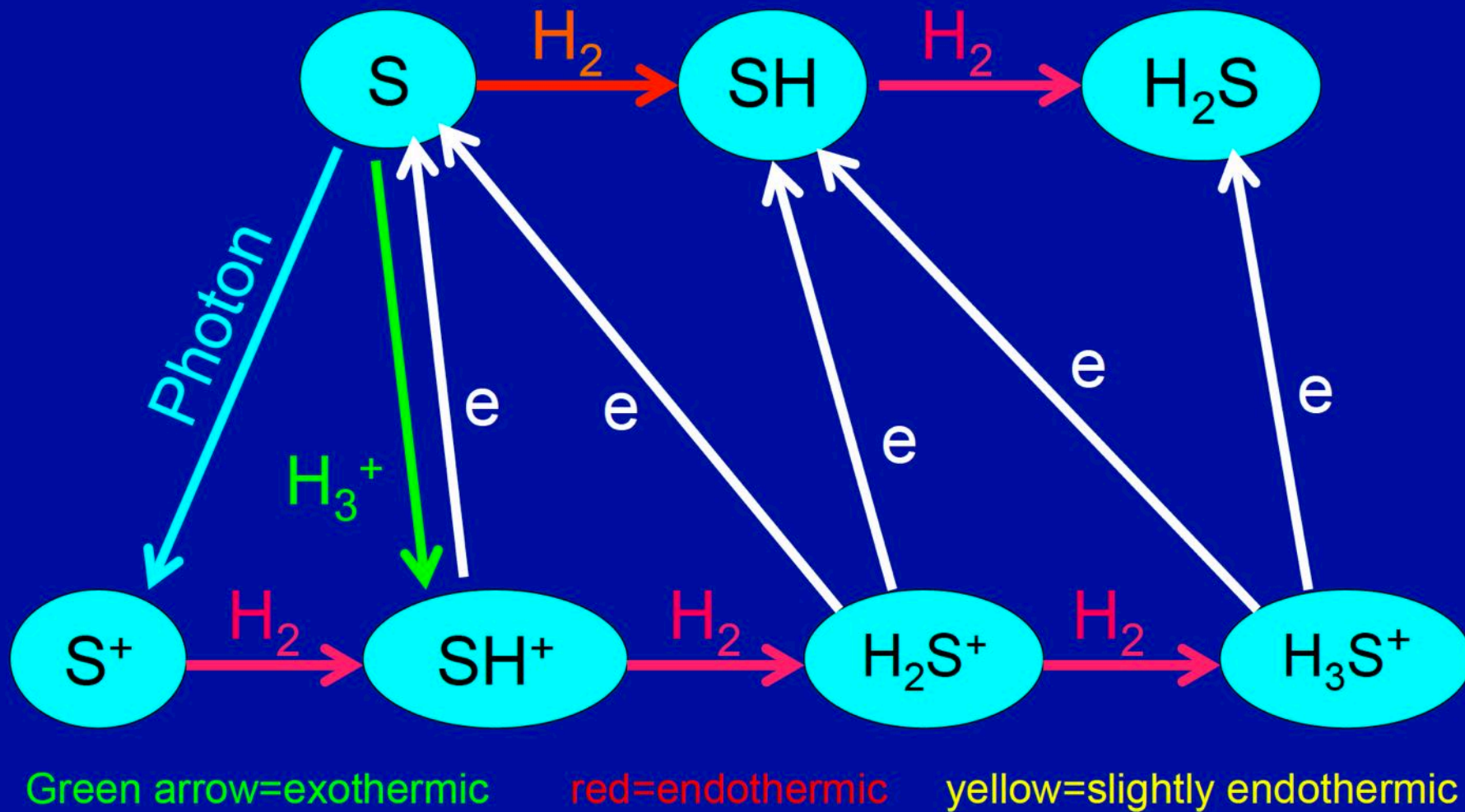
Neufeld et al. 2012, A&A



See also Neufeld et al. 2015

Underlying thermochemistry

SULFUR



Courtesy of David Neufeld

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Underlying thermochemistry

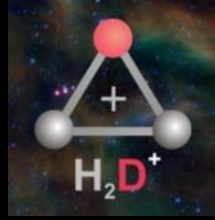
OH and H₂O can be produced via two pathways:

Low temperature: ion-molecule reactions, then dissociative recombination of H₃O⁺

High temperature: neutral-neutral reactions

SH and H₂S are only produced at elevated temperatures:

Their presence is evidence for shocks or turbulent dissipation regions



Summary



With **SOFIA** we can explore the foundations of astrochemistry:

- the deuteration of H_3^+ via HD – the starting point of D fractionation
- the ortho-to-para H_2 , via observations of para- H_2D^+ (and ortho- H_2D^+ with APEX) – important for D fractionation and cloud ages
- OH, OD (, carbonyl?) – oxygen chemistry and water fractionation – synergy with Herschel and APEX
- SH unveils the presence of shocks in diffuse clouds